

## ABSTRACT

Title of dissertation: *IN SITU* GROWTH AND DOPING  
STUDIES OF TOPOLOGICAL INSULATOR  
BISMUTH SELENIDE

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The past decade has borne witness to the rapid development of a new field of theoretical and experimental condensed matter physics commonly referred to as “topological insulators”. In a (experimentalist’s) single sentence a topological insulator can be thought of as material that behaves like an empty metal box: the  $i$ -dimensional material ( $i = 2,3$ ) is insulating, but conducting states exist at the  $(i-1)$ -dimensional boundaries. These edge or surface states possess non-trivial properties that have generated interest and excitement for their potential utility in solving practical problems in spin electronics as well as the creation of condensed matter systems for realizing and testing new physics. The most extensively studied materials systems with these properties suffer a major drawback in that the interior of the metal box is not “empty” (insulating) but instead filled with metal. The goal of this work has been to understand why the box is full instead of empty, and explore a particular pathway to making it empty.

To address these outstanding questions in the literature pertaining to the per-

sistent  $n$ -doping of topological insulator  $\text{Bi}_2\text{Se}_3$ , a custom apparatus was developed for combined epitaxial thin film growth with simultaneous, *in situ* measurement of transport characteristics (resistivity, Hall carrier density and mobility).  $\text{Bi}_2\text{Se}_3$  films are found to be  $n$ -doped before exposure to ambient conditions and this doping appears to be interfacial in origin. This work and methodology was extended to study the efficacy of an amorphous  $\text{MoO}_3$  capping layer.  $\text{MoO}_3$  acts as an electron acceptor,  $p$ -doping the  $\text{Bi}_2\text{Se}_3$  up to a  $|\Delta n| = 1 \times 10^{13} \text{ cm}^{-2}$  change in carrier density. Complimentary X-ray photoemission spectroscopy (XPS) on bulk crystals showed that this was enough to put the Fermi level into the topological regime. Thin films were too highly  $n$ -doped initially to reach the topological regime, but the same magnitude change in doping was observed via the Hall effect. Finally, a  $\text{Bi}_2\text{Se}_3$  film with a 100 nm capping layer of  $\text{MoO}_3$  was vented to ambient, and found to retain a stable doping for days of ambient exposure, demonstrating the effectiveness of  $\text{MoO}_3$  for passivation.

IN SITU GROWTH AND DOPING STUDIES OF  
TOPOLOGICAL INSULATOR BISMUTH SELENIDE

by

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## List of Abbreviations

AFM	Atomic Force Microscopy
HOMO	Highest Occupied Molecular Orbital
LUMO	Lowest Unoccupied Molecular Orbital
MBE	Molecular Beam Epitaxy
RHEED	Reflection High Energy Electron Diffraction
STI	Strong Topological Insulator
STM	Scanning Tunneling Microscopy
UHV	Ultra High Vacuum
WTI	Weak Topological Insulator
XPS	X-ray Photoemission Spectroscopy



## Chapter 1: Introduction

### 1.1 General Introduction

Topological Insulators are a recently discovered class of materials that have already been the subject of literature reviews from various subfield perspectives [18] [19] [20] [21] [22], as well as a (now dated, but) definitive and comprehensive review in *Reviews of Modern Physics* [23]. The purpose of this chapter is to give a brief overview of the origins of the field, and subsequently focus on current literature concerning bulk doping and surface impurities in these materials, which will be relevant to the subsequent chapters of this thesis.

The field of condensed matter physics has long been defined by Landau's classification of matter and phase transitions according to symmetry breaking. The most accessible example of this is the phase transition of water from liquid to ice. Liquid water's constituent molecules, while highly correlated, lack long range order: liquid water has continuous translational and rotational symmetry. Ice is crystalline, conforming to a lattice, so translational symmetry becomes discrete instead of continuous, and rotational symmetry is broken. In magnetism, the Curie temperature can be described in terms of breaking rotational symmetry (paramagnetic states are rotationally symmetric; Ferro-, Ferri-, and Antiferromagnetic states aren't). The

most simple superconducting transition is described by the energy gap order parameter, understood in terms of gauge symmetry breaking. Despite this breadth in applicability of symmetry breaking, the beginning of the 1980's saw the discovery of the quantum Hall effect, experimental evidence that symmetry breaking was not a universally applicable explanation for distinguishing new phases.

The (integer) quantum Hall effect is observed in a two dimensional electron gas subjected to an out of plane magnetic field. The effect of the magnetic field is to force the electrons into cyclotron orbits at discrete values of energy called Landau levels. This leads to the well known quantization of Hall conductivity:

$$\sigma = \frac{e^2}{h} \nu; \nu \in \mathbb{Z} \tag{1.1}$$

Initially observed by von Klitzing in 1980 [24], figure (1.1) shows a more illustrative experimental realization of the plateaus caused by the quantization of the transverse conductivity.

What is clear from the figure is that the system is transitioning between distinct, measurably different states described by different integers  $\nu$ , but in this case there is no simple explanation using symmetry breaking. Instead, the language and mathematics of topological order has been used to explain these transition phenomena.

The first explanation of this effect was offered by Laughlin, who analyzed the problem using the magnetic flux through a cylindrical ring geometry [25]. Limitations to this approach were pointed out by Thouless, Kohmoto, Nightingale and

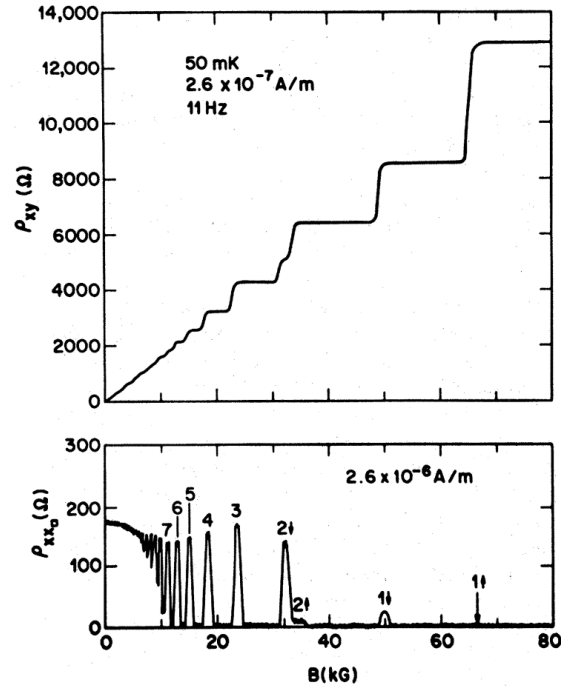


Figure 1.1: Transverse ( $\rho_{xy}$ ) and longitudinal ( $\rho_{xx}$ ) resistances measured against perpendicular magnetic field in a GaAs- $\text{Al}_x\text{Ga}_{1-x}\text{As}$  2DEG, demonstrating the integer quantum Hall effect. Figure reproduced from reference [1].

den Nijs (TKNN), who used the Kubo linear response formula on a periodic torus surface to derive the same quantized conductance [26]. The phase space geometry of the TKNN result in particular is crucial to making the connection to the underlying mathematics.

The Gauss-Bonnet theorem from the field of differential geometry connects the integral of the Gaussian curvature ( $K$ ) over a closed two-dimensional surface ( $S$ ) to the Euler characteristic of the surface:

$$\frac{1}{2\pi} \int_S K dA = \chi(S) \in \mathbb{Z} \quad (1.2)$$

Chern showed that this result could be generalized to a broader class of manifold [27]. This result allowed for the Gauss-Bonnet theorem to be applied instead to the integration over the Brillouin zone of the ‘curvature’ of wavefunctions, which is better known as the Berry curvature (the bracketed quantity in eq. 1.3) [28]:

$$\frac{i}{2\pi} \int_{BZ} d^2k [\nabla \times \langle u_m | \nabla_k | u_m \rangle] = n_m \in \mathbb{Z} \quad (1.3)$$

Here the  $m$  subscript refers to the particular band, and the  $|u_m\rangle$  refer to the Bloch wavefunctions. The integer in this case is referred to as the Chern number for the  $m^{th}$  band. Shortly after the TKNN result was derived, Avron *et al.* identified the underlying mathematics of their method as a particular Chern class [29], [30]. In a recent review of the field [18], Ando provided a simple calculation using perturbation theory to derive the quantized Hall conductivity in a 2D system of electrons, equation (1.1), where he reduces his expression for the Hall conductivity  $\sigma_{xy} = \frac{\langle j_x \rangle}{E_y}$

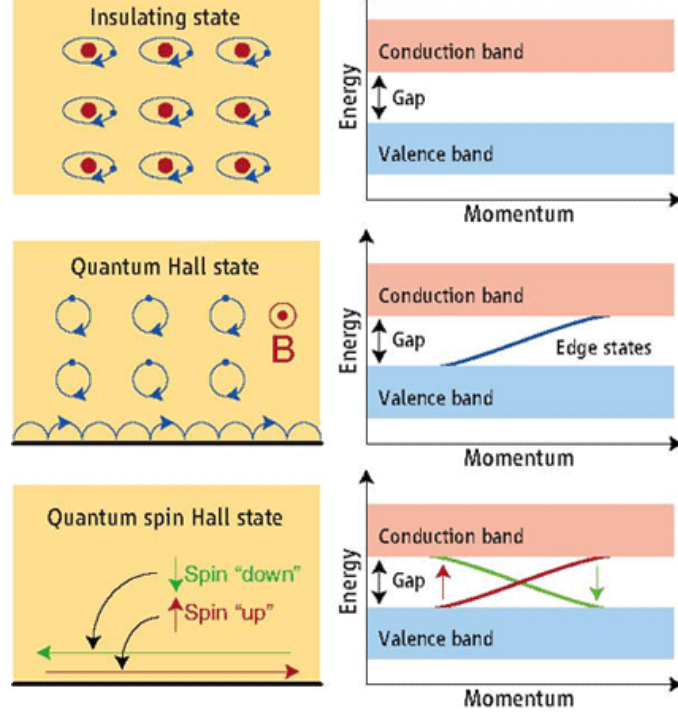


Figure 1.2: Illustration of the distinction between quantum and spin quantum Hall effects. The left hand images are in real space, the right hand images show the corresponding dispersion relations. Figure reproduced from references [2], [3].

(current density at right angle to the  $E$ -field direction) to an expression involving the Berry phase:

$$\sigma_{xy} = \frac{e^2}{h} \sum_n \frac{i}{2\pi} \int_{BZ} d^2k [\nabla \times \langle u_{nk} | \nabla_k | u_{nk} \rangle]$$

where the summation in  $n$  counts the contributing bands, and the expression integrated over the Brillouin zone is exactly equation (1.3), and therefore an integer. In this simple model of a square 2D electron system of finite size, with an applied out of plane magnetic field and in plane electric field, the Hall conductance is quantized in multiples of  $e^2/h$ : in this case the topological mathematics of the Chern number can be thought of as ‘counting’ the number of contributing bands.

The topological description of the quantum Hall effect laid the groundwork

for the theoretical prediction of the so-called quantum spin Hall effect [31] [32] [33]. While the Hall effect is the spatial separation of different charges due to a magnetic field, the spin Hall effect is the momentum separation of different spin states as the result of application of an electric field. This difference is highlighted in figure 1.2, which illustrates the difference between three different states. In an insulating state, electrons are highly localized, which manifests itself as the presence of a gap in the band structure. The quantum Hall state is characterized by the presence of a magnetic field causing the electrons to “run loops”, where the loop size is quantized. This motivates a simple way to think about the presence of an edge state, where the loop-like path of the electron repeatedly runs into the boundary of the conduction media, imposing a momentum direction. The quantum spin Hall effect comes from a spin orbit interaction term instead of an external magnetic field. This means that electrons with opposite spin states have to be traveling in opposite directions, which leads to unusual boundary phenomena: these phenomena are why topological insulators are so interesting.

The advantage of using the aforementioned Chern classification scheme is that it can be expanded to quantitatively describe the more complicated quantum spin Hall edge states. Kane and Mele extended the Chern class idea by introducing a so called  $Z_2$  topological invariant [34]. To get to the  $Z_2$  invariant, it is instructive to start with Haldane’s model for graphene [4].

He considered in particular a periodic local magnetic flux density “with the full symmetry of the lattice, and with zero total flux through the unit cell”, visualized in detail in figure (1.3). He showed that despite no net magnetization, it was possible to

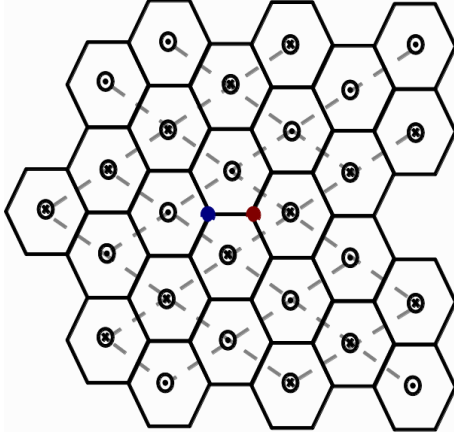


Figure 1.3: Illustration of the Haldane model for graphene. The grey dotted lines delineate the unit cells of graphene (enclosing the two sublattices- blue and red dots). The dots and crosses represent the periodic placement of magnetic dipole moments leading to zero total flux through each unit cell. Figure created based on the description in reference [4].

break time reversal symmetry, which leads to an integer quantum Hall state, instead of just a trivial semiconductor. Kane and Mele's essential innovation [34] was to expand the Haldane model in a time reversal symmetric manner by including spin-orbit interaction terms, as a way of accounting for spin. They considered the time reversal operator on a two band model using mathematics common in conformal field theory, constructing an integer result to a contour integration in the Brillouin zone whose result (odd or even) determines if the system in question exhibits the quantum spin Hall phase (odd) or is a trivial insulator (even)- the  $Z_2$  invariant. They further studied spin orbit coupling in graphene to see if the quantum spin Hall effect might be experimentally accessible [35]; this was subsequently shown to be unrealistic because the effect would be too weak [36]. An alternate route to the  $Z_2$  index was given by Xu and Moore [37], a many-body approach exploiting the time reversal symmetry condition and Kramers degeneracy, allowing them to investigate

the stability of the quantum spin Hall state to interactions and disorder. Bernevig and Zhang also studied this problem, using a strain gradient to induce quantization, strictly preserving time reversal symmetry [33]. They extended this work to make a concrete theoretical prediction for finding this state in HgTe quantum wells [38], experimentally verified thereafter [39].

Figure 1.2 provides a simple illustration highlighting the similarities and differences between the quantum, and quantum spin Hall effects. In both instances one sees edge states produced by an external field acting on the electrons. However, here the distinction of the time reversal symmetry become manifest: the QSHE is symmetric in momentum about the edge state crossing point, where no such symmetry exists in the QHE band diagram. One can also see that the spin states are momentum segregated in the QSH state, no such distinction being present in the QH state.

This discussion has so far been limited to two dimensions. However, the Kane and Mele [34] description of the underlying topological invariants allowed for generalizations of  $Z_2$  into higher dimension. Kane and Mele's original work on  $Z_2$  was first extended to three dimensions by Moore and Balents, who used homotopy theory to show that in 3D, each band pair is characterized by four  $Z_2$  integers instead of just one [40]. Building on this work, Fu, Kane and Mele offered a different approach to the four  $Z_2$  integers  $\nu_0;(\nu_1\nu_2\nu_3)$ , and an interpretation of the latter three as related to the Miller indices of the reciprocal lattice space [5]. The  $Z_2$  integers are calculated relative to the time reversal invariant momentum (TRIM) points from the reciprocal lattice vectors. They enumerate the different ways a Fermi arc can



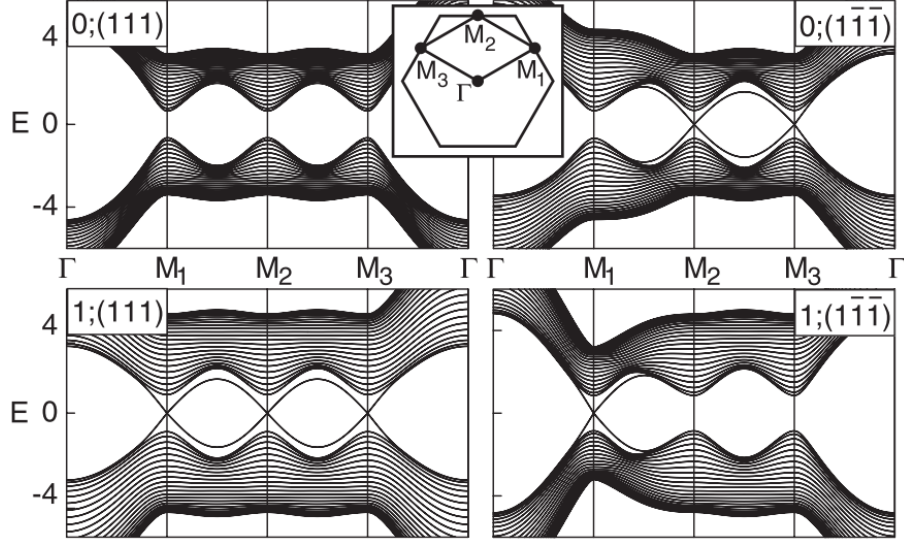


Figure 1.4: Band structure calculation for a (111) slab of diamond lattice. The four subplots are different phases labelled according to their Miller indices. In the top (bottom) two diagrams  $\nu_0=0$  (1), and there are an even (odd) number of crossings at  $E=0$ . Figure reproduced from [5].

enclose the TRIM points in the Brillouin zone. The parity of the first index  $\nu_0$  defines two broad categories:  $\nu_0 = 0$  are weak topological insulators (WTI), where the consequent  $\nu_{1,2,3}$  indices force an even number of Dirac points enclosed by the surface Fermi arc. This makes these surface states susceptible to localization as a result of disorder. The strong topological insulator (STI) case is when  $\nu_0 = 1$ : an odd number of Dirac points are enclosed by the surface Fermi arc. This introduces a  $\pi$  Berry phase to the electron wave function, suppressing back-scattering [41], the phenomena commonly referred to as “topological protection” and assigned in real space to the surface. This time-reversal symmetric 3D topological insulator state will be the major topic of this thesis and I will refer to it generically as “topological insulator” in the subsequent chapters.

They employed this analysis to consider a tight binding model of a diamond

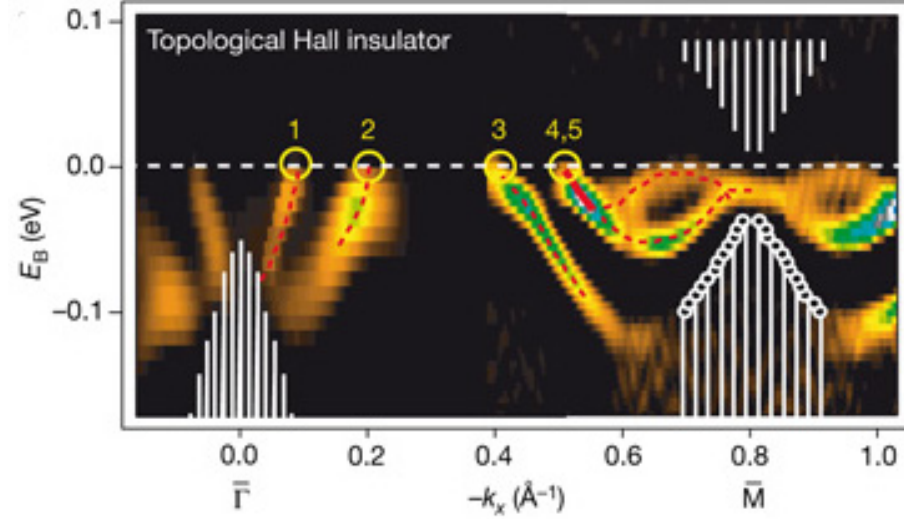


Figure 1.5: ARPES measurement of  $\text{Bi}_{0.9}\text{Sb}_{0.1}$  along  $\bar{\Gamma} - \bar{M}$ . The white lines are a projection of bulk band data, the yellow circles indicate the (odd) number of surface state Fermi level crossings. Figure reproduced from reference [6].

lattice with the spin-orbit interaction, and saw these different (weak, strong) phases in different crystallographic directions. Figure 1.4 shows their band structure computations for different sets of topological indices, where this odd/even paradigm is explicitly manifest in the number of Dirac cone band crossings at  $E=0$ .

Fu *et al.* [5] also built a tight binding model and applied their results to bismuth, in consideration of Murakami’s work showing that bismuth would be a good candidate material in which to try and realize the quantum spin Hall state [42]. Further work by Fu, Teo, and Kane tightened theoretical predictions for the specific alloy system  $\text{Bi}_{1-x}\text{Sb}_x$  [43], [44]. This was within months of the first experimental work on the same material system, where angle resolved photoemission spectroscopy (ARPES) showed an odd number of Fermi level crossings providing evidence for the topological nature of the material (figure 1.5) [6]. ARPES utilizes a source of high intensity photons (e.g. a synchrotron) to photoexcite electrons in a given sample;

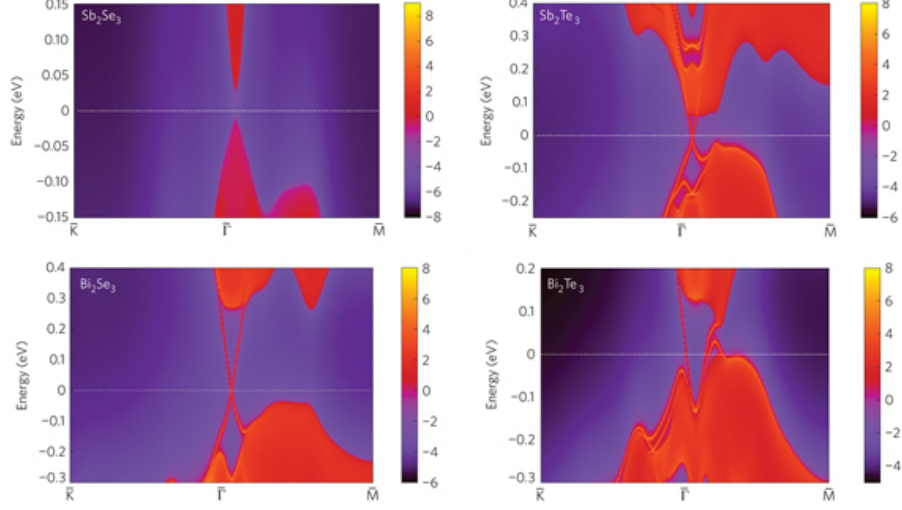


Figure 1.6: Energy and momentum dependence for the denoted chalcogenide systems calculated theoretically using density function theory. The red color indicate a higher local density of states, the presence of a band. Figure reproduced from reference [7].

measuring the momentum of these electrons directly maps the band structure ( $E$  vs.  $k$ ). This means that figures 1.5 and 1.4 are plotting the same things, albeit for different material systems, and experiment vs. theory.

However,  $\text{Bi}_{1-x}\text{Sb}_x$  is an alloy. This means there isn't an easily defined crystalline structure, making it hard to model theoretically. For this reason, Zhang et al. chose instead to investigate the layered stoichiometric crystals  $\text{Sb}_2\text{Te}_3$ ,  $\text{Sb}_2\text{Se}_3$ ,  $\text{Bi}_2\text{Te}_3$ , and  $\text{Bi}_2\text{Se}_3$  (figure 1.6) [7].

Their first-principles calculations predicted both a single Dirac cone crossing and a bulk gap of 0.3 eV for  $\text{Bi}_2\text{Se}_3$ , notable for how it is an order of magnitude larger than the energy scale of room temperature, 0.025 eV. This was experimentally confirmed by Hasan's group, who had moved to the  $\text{Bi}_2\text{Se}_3$  system from  $\text{Bi}_{1-x}\text{Sb}_x$  for the same reasons [8]. They grew single crystal of  $\text{Bi}_2\text{Se}_3$  via the Bridgman

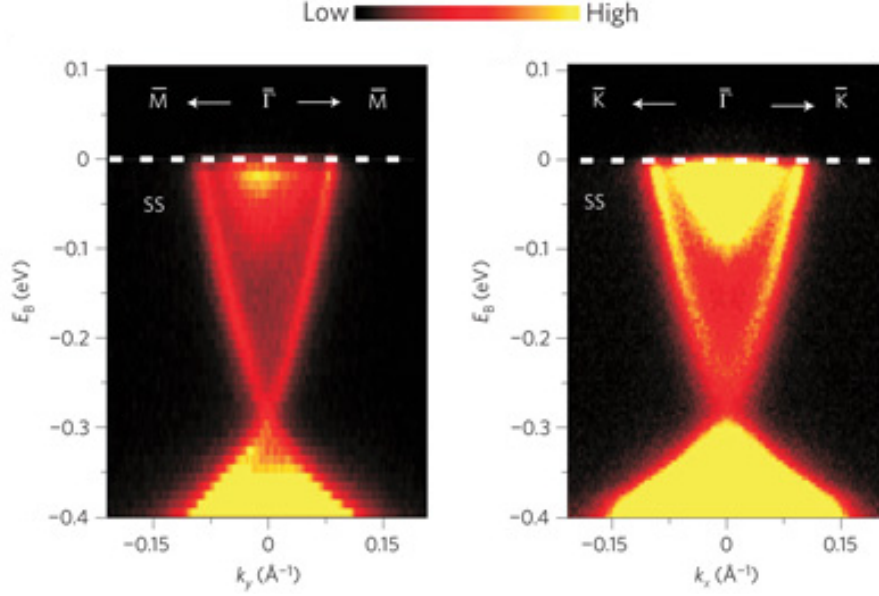


Figure 1.7: First experimental ARPES measurement of surface electronic band dispersion for cleaved  $\text{Bi}_2\text{Se}_3$  single crystal, taken along two different momentum-space cuts. The bright red 'Dirac cone' is evidence of the surface state for example predicted in the lower left of figure 1.6. Note the position of the Fermi level ( $0 E_B$ ) in the bulk conduction band. Figure reproduced from reference [8].

method, which was subsequently cleaved in vacuum and characterized by ARPES (figure 1.7).

The spin-correlated 'Dirac'-like dispersion of the surface states is the core concept of topological insulators. Expanding the nuance and complexity of our understanding of condensed matter systems has intrinsic value, in the spirit of Anderson's manifesto 'More is Different' [45], but has generated additional excitement for the regions of previously unexplored physics these systems could be used as a building block to access. Qi and Zhang do a good job outlining in broad strokes some of the potential applications, notably heterostructures that would act like magnetic monopoles, or exhibit non-Abelian statistics [2]. A less esoteric application would be the obvious utility of control of spin-correlated surface conduction states in de-

veloping practical spin electronics, a possibility broadly outlined by Pesin and MacDonald [22]. These exciting avenues of new physics and applications have generated considerable excitement, but formidable obstacles remain before their realization. The data in figure (1.7) clearly demonstrates an immediate experimental shortcoming that is the subject of active research. As noted in the caption, the position of the Fermi level in the system is far into the bulk conduction band, making this sample of  $\text{Bi}_2\text{Se}_3$  a bulk conductor, instead of a bulk insulator. These bulk states in practice make the surface states of interest incredibly hard to detect in any experimental transport scheme, a necessary precursor for exploiting their unusual properties. The following sections review various approaches to dealing with this problem.

## 1.2 Experimental Approaches to Access Topological Surface States

There are several experimental routes to making surface state phenomena more prominent in a transport measurement, among them chemical doping to move the Fermi level, and designing the geometry of the device to increase the ratio of surface to bulk contribution. The drawback to chemical doping is the disorder into the system it necessarily introduces. Preparation methods that allow the control of surface/bulk ratio (thin film or nanowire growth) generally produce materials of much lower quality than bulk grown crystals. For the purposes of accessing the surface states, there are also various schemes of electrostatically tuning the Fermi surface into the bulk gap of the chalcogenide material systems.

This series of bismuth-based chalcogenides have been studied extensively in

the past for their thermoelectric properties, so their crystal structure [46] and bulk characteristics [47] have been known for decades, while the ‘surface states’ have only been subject to much more recent scrutiny. As recently as 2002,  $\text{Bi}_2\text{Se}_3$  was being discussed in terms of its thermoelectric properties, establishing from ARPES data that bulk crystals are  $n$ -doped [48], also noted by Xia *et al.* [8]. This issue of doping has been studied extensively from the perspective of optimizing thermoelectric material systems, in particular  $n$ -type  $\text{Bi}_2\text{Te}_3$  [49].

The crystal structure of these chalcogenides is notable in having quintuple layers of alternating bismuth and selenium or tellurium that are strongly bonded to one another (figure 1.8). However, between the quintuple layers, the bonding is of the weaker van der Waals type [7]. This makes the bulk crystal resemble graphene in that the ‘scotch tape’ method pioneered by Geim’s group to exfoliate thin layers of graphite [50] can be applied to make similarly thin flakes and therefore potentially reduce the bulk contribution to transport. In addition to bulk exfoliation, it is also possible to grow these materials by molecular beam epitaxy (MBE) on a variety of substrates [51] [52] [53]. This process is referred to as “van der Waals epitaxy” because of the inter-quintuple-layer bonding; this particular bonding allows films of these materials to be grown on substrates with a large lattice mismatch.

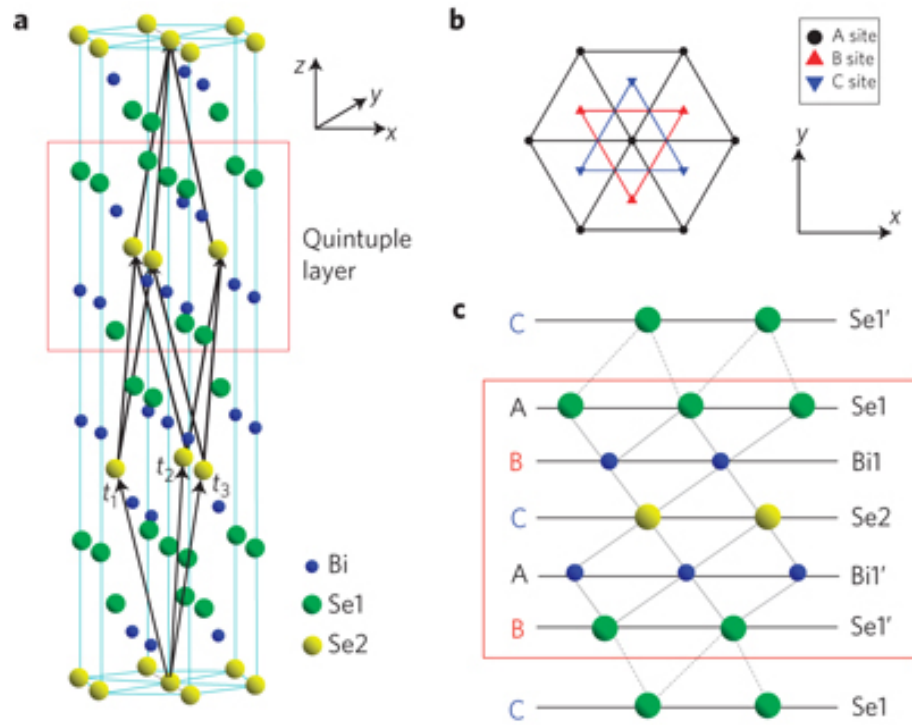


Figure 1.8:  $\text{Bi}_2\text{Se}_3$  crystal structure, with additional side view of the stacking order. Figure reproduced from reference [7].

## 1.2.1 Chemical Mediation for Bulk Suppression

### 1.2.1.1 Se Vacancies

Selenium vacancies are universally cited as one of the leading reasons for the persistent  $n$ - doping of even the most meticulously prepared bulk crystals of  $\text{Bi}_2\text{Se}_3$  [54]. This has been confirmed by direct STM/STS measurement of Se defect sites (see for example [55]). Various other approaches beyond preparation of the most stoichiometric  $\text{Bi}_2\text{Se}_3$  possible are also being extensively researched.

### 1.2.1.2 Chemical Substitution

One approach to addressing the intrinsic doping problem of bismuth selenide involves substitutional doping. Initial efforts focused on calcium doping of  $\text{Bi}_2\text{Se}_3$  to compensate for the electron doping produced by selenium vacancies (figure 1.9, [56] [9]).

Devices fabricated from similarly Ca- compensated bulk  $\text{Bi}_2\text{Se}_3$  have shown evidence of having sufficiently low initial bulk carrier density to be successfully gated into the surface state dominant regime [57]. Pb and Sb have also been shown to successfully move the Fermi surface from above to below the Dirac point [58].

### 1.2.1.3 Stoichiometries Beyond Binary Compounds

More recent work has focussed on even more complicated stoichiometries involving Bi, Sb, Se, and Te (see for example [59], [60], [61] [62]). A review of ternary



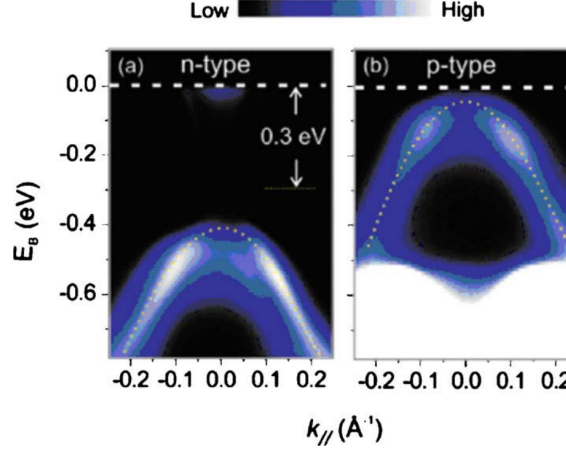


Figure 1.9: ARPES data comparing  $n$ -type  $\text{Bi}_2\text{Se}_3$  (left) with  $\text{Bi}_{1.98}\text{Ca}_{0.02}\text{Se}_3$  (right), showing a clear change in the position of the bulk bands. The explanation in the paper leaves ambiguous whether only observing the bulk bands (no evidence of the surface state in the gap) was deliberate. Figure reproduced from reference [9].

compounds beyond just the bismuth based materials has recently been completely by the Hasan group [63], [64]. One of the most recent transport works available on the  $\text{Bi}_2\text{Te}_2\text{Se}$  system identifies the crossover temperature from bulk to surface conduction at 30K [65], which is still a far way off from ready electronic access at room temperature.

In the so-called BSTS system, the specific stoichiometry of  $\text{Bi}_{1.5}\text{Sb}_{0.5}\text{Te}_{1.7}\text{Se}_{1.3}$  has very promising transport characteristics. Lee *et al.* calculate that their samples have a bulk:surface conductance ratio of 26% (6%) at 290 K (4.2 K) [66]. In more recent work using crystals of  $\text{Bi}_{1.5}\text{Sb}_{0.5}\text{Te}_{1.8}\text{Se}_{1.2}$ , Wang *et al.* reported an 8% surface conductance contribution at low temperature, where they analyzed their data with respect to surface, bulk and impurity bands [62]. STM [67] and ARPES [68] studies provide evidence that the Fermi level can be tuned to the Dirac point with the correct composition.

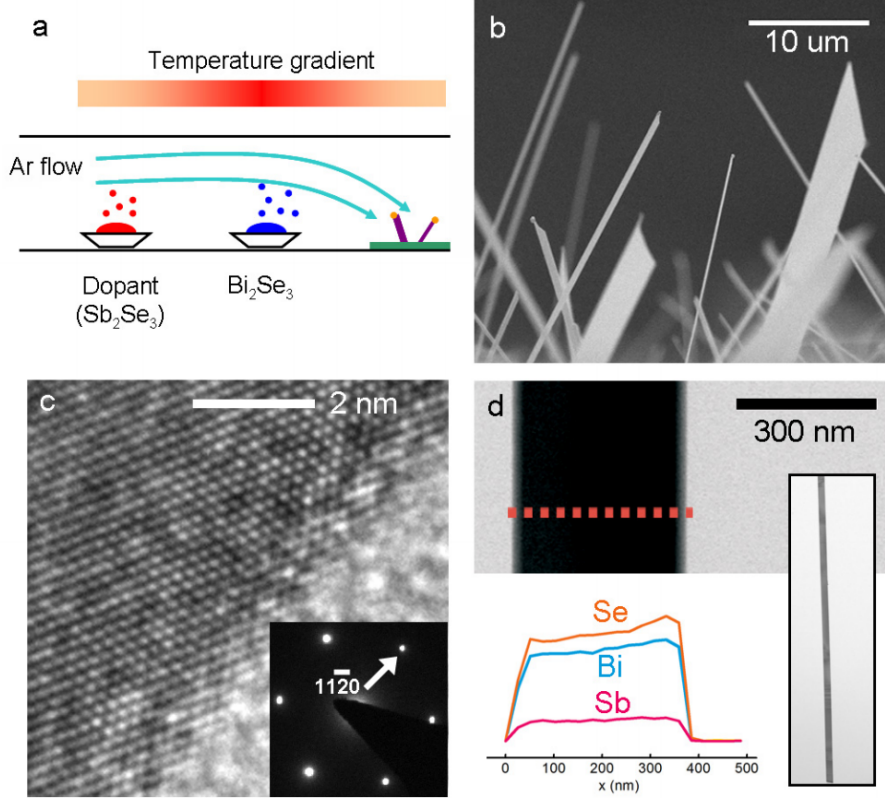


Figure 1.10: Schematic showing vapor-liquid-solid growth technique, and resultant nanoribbons of  $(\text{Bi}_{1-x}\text{Sb}_x)_2\text{Se}_3$ , characterized by SEM (b) and TEM (c). d) is an EDX profile showing the consistency of dopant introduced. Figure reproduced from reference [10].

### 1.2.2 Geometric Suppression of Bulk States

Nanoribbons by construction favor surface effects over the bulk, because of the favorable ratio of surface area to volume. Various efforts have shown success in using a vapor-liquid-solid growth technique to grow nanoribbons of  $\text{Bi}_2\text{Se}_3$  (figure 1.10) [69] [10].

Hong *et al.* [10] showed that subsequent Sb doping combined with a ZnO protective layer allowed them to access the ambipolar transport region associated

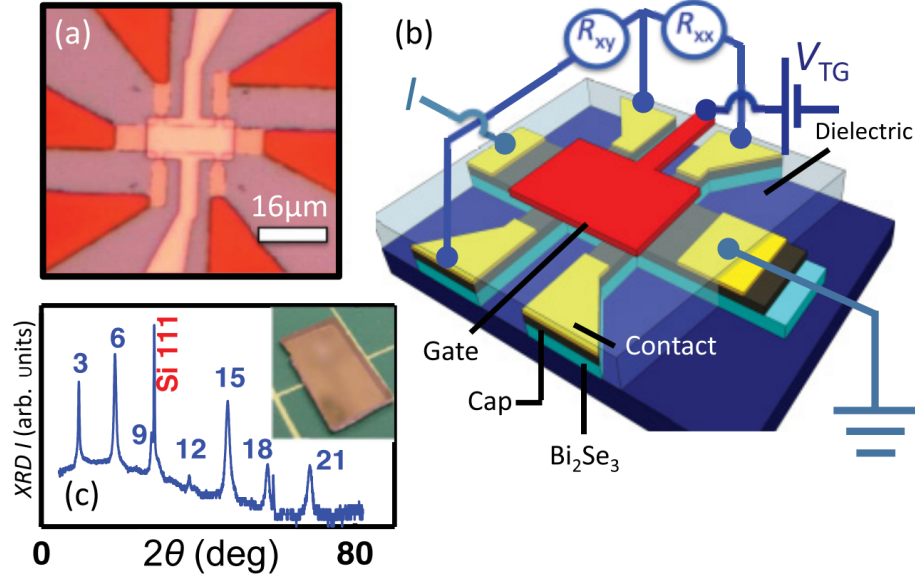


Figure 1.11: a) shows an optical image of the particular photopatterned device with contacts and a top gate. b) shows the general scheme for top and bottom gating. c) is X-ray diffraction data giving a measure of the quality of the Bi<sub>2</sub>Se<sub>3</sub> film, indicating that it is growing in the c-axis direction, matching the substrate termination used. Figure reproduced from reference [11].

with surface transport [12].

### 1.2.3 Electrostatic Gating

There now exists a host of literature using both bulk exfoliated and thin film devices with various gating schemes to quantify the transport properties of these materials. Photopatterned devices of bismuth selenide and telluride have been gated from the top [11] [70], and also through the substrate [71]. Figure (1.11) is broadly representative of the general device geometry used. Figure (1.12) shows representative transport evidence for successful gating from *n*-doped to *p*-doped, through the bulk gap.

Band bending has been employed in a variety of experimental studies to de-

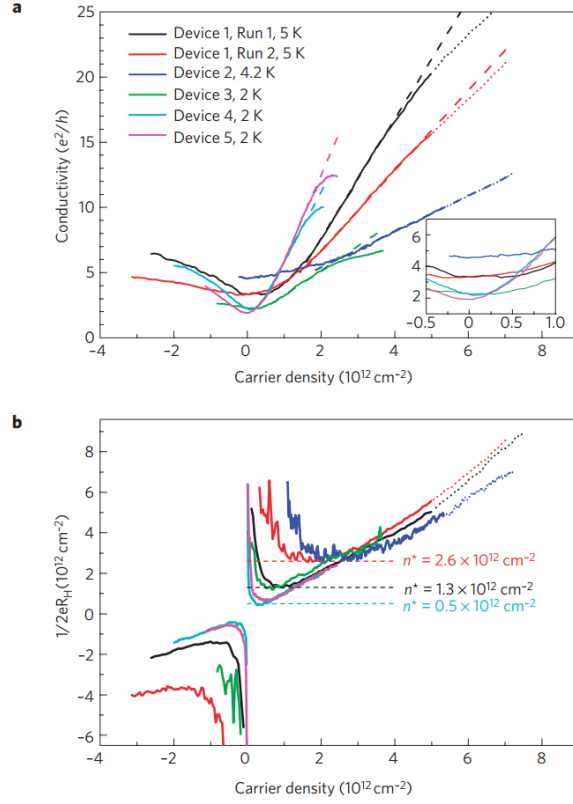


Figure 1.12: Exfoliated  $\text{Bi}_2\text{Se}_3$ , gated with an electrochemical double layer capacitor scheme. a) shows the conductivity minimum (Dirac point) achieved in a variety of samples as a function of the carrier density, extracted from Hall data. b) shows the change in sign of the charge carriers, indicating a transition from  $n$ - to  $p$ -doped material. Here  $n^*$  is the residual carrier density, attributed to electron and hole puddles. Figure reproduced from reference [12].

scribe the various conduction schemes involving the bulk, top and bottom surfaces [72] [11] [73] [10] [57]. It has also been used theoretically to connect the initial  $n$ -doping of these materials with their subsequent poor bulk resistivity [74]. Checkelsky *et al.* attribute a reduction in the size of the bulk band gap to band bending, with negative implications for accessing the surface states at room temperature [57]. Steinberg *et al.* lay out a very careful argument using interfacial band bending to explain where the Fermi surface sits at the top and bottom surfaces in their exfoliated, top and bottom gated flakes of  $\text{Bi}_2\text{Se}_3$  [73]. However, this is hard to reconcile with recent results from Kim *et al.*, who showed no difference between the effects of back or top gate, suggesting that the surfaces are electrostatically coupled (difference in thickness between the two studies) [12]. Band bending could still be relevant in the presence of charge traps in the bulk, or large dopant compensation, due to the amounts of charge that are being electrostatically compensated. It could also be the case that the temperature dependence interpreted as activation from bulk valence to surface conduction bands in the Kim study was incorrectly attributed to band bending in refs. [57] and [73]. This is speculative at this time and remains an area of active research.

### 1.3 Magnetic Impurities

The spin-orbit interaction is time reversal symmetric. The degeneracy at the Dirac point is protected by time reversal symmetry: the two states are a Kramers pair thus must be degenerate as long as there is time reversal symmetry. Introducing

magnetic impurities would break time reversal symmetry in a localized manner, in particular affecting the surface states of the topological insulator. Initial theory work demonstrated magnetic impurities suppressing the local density of states, opening a local gap [75]. Subsequent theoretical work is more material specific, for example first principles calculations for Co impurities on  $\text{Bi}_2\text{Se}_3$  [76], or Mn doping of  $\text{Bi}_2\text{Te}_3$  [77]. Schmidt *et al.* conclude that Co doping opens a gap [76], while Henk *et al.* compute a non-trivial surface spin texture, and survival of the surface state [77]. This same diversity of results is present in the experimental literature on doping with magnetic impurities. Both magnetic and nonmagnetic impurities (Cs, Gd, Rb) deposited on bulk  $\text{Bi}_2\text{Se}_3$  with concurrent ARPES, showed identical, negligible degradation of the surface state (figure 1.13) [13].

A substantial number of studies done using Fe as a dopant have also yielded conflicting conclusions. Studies of Fe deposited on the surface of  $\text{Bi}_2\text{Se}_3$  have been interpreted as evidence of opening a band gap [78] [79], and lack of a band gap [80]. A later STM study showed an in-plane magnetic anisotropy for Fe impurities, evidence that a gap is not opened [81]. X-ray magnetic circular dichroism of Fe and Co deposited on bulk cleaved  $\text{Bi}_2\text{Te}_3$  concluded that the adatoms do not induce long range magnetic order [82]. This conclusion is supported by another STM work done with Co doping [83].

In contrast with all the surface dopant studies demonstrating how magnetic dopants do not destroy the surface states, thin films of  $\text{Bi}_2\text{Se}_3$  grown with increasing concentrations of Cr [84], and Mn [85], show a transition from weak anti-localization to weak localization, with additional ARPES measurements supporting

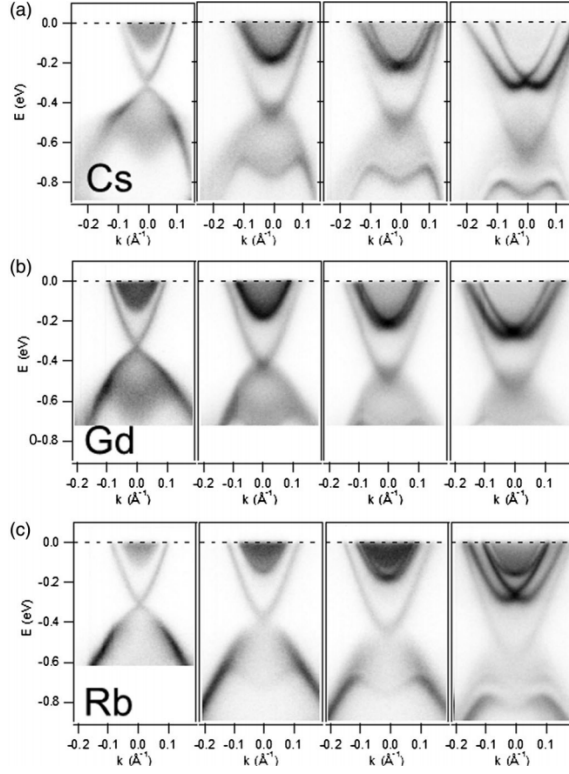


Figure 1.13: Cleaved, bulk  $\text{Bi}_2\text{Se}_3$  with Cesium (a), Gadolinium (b) and Rubidium (c) adsorped on the surface. From left to right, pristine to maximally doped surface. Note the persistence of the surface bands despite increased doping. Figure reproduced from reference [13].

the conclusion that the surface states are suppressed. Similar work on Mn doped  $\text{Bi}_2\text{Te}_{3-y}\text{Se}_y$  showed results suggesting one-dimensional conduction along magnetic domain walls [86], consistent with the picture that magnetic impurities will create local ferromagnetic domains without inducing long range ferromagnetic order.

## 1.4 Nonmagnetic Impurities

Another theoretical consequence of the  $\pi$  Berry phase and spin texture of topological insulator surface states is the predicted suppression of backscattering from nonmagnetic impurities on the surface. Evidence of this was inferred from the standing wave patterns observed by low temperature STM imaging of Ag impurities on the surface of  $\text{Bi}_2\text{Te}_3$  film [87]. Similar work on  $\text{Bi}_2\text{Se}_3$  showed intercalation of the Ag impurities into the quintuple layers, but ARPES results indicated that the surface states were preserved [88]. Ye *et al.* noted that while backscattering is theoretically prohibited under conditions of time reversal symmetry, scattering from arbitrary angles is not necessarily prohibited. Their work argues that the intercalation of Ag impurities pushes the real space location of the surface state into the bulk, away from the physical surface, which may provide an avenue for preserving its unique properties despite environmental exposure.

## 1.5 Environmental Effects

A great deal of work has been done characterizing the response of the chalcogenides to environmental exposure, work directly related to the doping problems



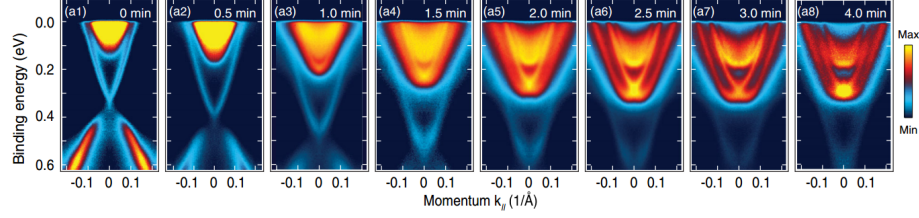


Figure 1.14: ARPES data of bulk cleaved  $\text{Bi}_2\text{Se}_3$  systematically exposed to Potassium. Evidence of Rashba splitting in the more heavily doped spectrum (degeneracy lifted in momentum), and shift of the Fermi level further into the bulk conduction band. Figure reproduced from reference [14].

previously discussed in the context of transport studies. Degradation of the surface state attributed to oxidation has been measured in  $\text{Bi}_2\text{Se}_3$  using x-ray photoelectron spectroscopy (XPS) [89]. An exfoliated bulk crystal of  $\text{Bi}_2\text{Se}_3$  systematically exposed to trace amounts of water was demonstrated by ARPES to develop quantum well states at the surface, attributed to band bending [90]. This same effect was also observed in the  $\text{Bi}_2(\text{Se}_{3-x}\text{Te}_x)$  system [91]. When potassium was used instead of water, the result was similar quantum well states developing on the surface, as well as a downward shift of the Dirac point (figure 1.14) [14]. This change in doping puts the Fermi level further into the bulk conduction band, away from the ‘topological regime’ where the surface states would be predominant in transport.

In a different study, a film of  $\text{Bi}_2\text{Se}_3$  was exposed to atmosphere for days, and repeatedly measured. The magnetic field dependence was fit to a two-carrier model (surface and bulk carriers) [92]. They interpreted their measured increase in doping and decrease in mobility observed on exposure to atmosphere as predominately in the bulk state, and that the surface doping and mobility remained relatively stable. This supports an interpretation of the surface states being robust, but the bulk of

the material being acutely sensitive to environmental exposure. DFT focused on surface oxidation concluded that chemisorption of  $O_2$  does not affect the existing surface states, where reaction with atomic O does effect a change [93]. There is a consensus in the literature that the particular chemistry associated with exposure to atmosphere (e.g. water and oxygen) is detrimental to achieving a low doping to make the surface states easily accessible in devices.

In work similar to the aforementioned studies done with Ag ([87] [88]), Rb deposited on  $Bi_2Se_3$  measured via STM demonstrated evidence that Rb becomes intercalated upon annealing; subsequent ARPES indicates stability of bands to systematic oxygen exposure [94]. Bianchi *et al.* [94] have a similar conclusion to Ye *et al.* [88] about the feasibility of using intercalation to gain better control of the surface states.

### 1.5.1 Selenium Capping Layer

Another possible avenue to controlling the surface states, or at least minimizing the environmental effects, is a capping layer. Lang *et al.* grew  $Bi_2Se_3$  film by MBE [95], with Shubnikov deHaas oscillations providing evidence that the presence of an *in situ* 2nm Al passivation layer on top of the TI film preserved the surface state of interest. As mentioned before, Hong *et al.* used a ZnO layer to protect their  $Bi_2Se_3$  nanoribbons [10].

Another common method, especially with thin film growth, is to add a Se capping layer to protect the surface of the film [96] [84]. The ability to de-cap and

restore a pristine  $\text{Bi}_2\text{Se}_3$  surface has been demonstrated via STM [97], and has been used in various studies involving bismuth selenide films (see for example [98], [99]).

## 1.6 Surface Transfer Doping

Surface transfer doping is another potential method to remove the *n*-type doping. It provides an alternative route to conventional bulk doping, as it does not involve introducing impurities into the bulk lattice and has been successfully implemented across a range of other materials [100] [101] [102] [103]. In the case of acceptor doping, high electron affinity molecules deposited onto the surface result in the transfer of electrons from the substrate into the molecular overlayer. This form of molecular doping on  $\text{Bi}_2\text{Se}_3$  was recently demonstrated using the strong electron accepting molecule tetrafluorotetracyanoquinodimethane (F4-TCNQ) on mechanically exfoliated thin  $\text{Bi}_2\text{Se}_3$  crystals on  $\text{SiO}_2/\text{Si}$  substrates. It was found that enough excess charge was removed to place the Fermi level near the conduction band edge, allowing the field-effect to be utilized to further tune the Fermi level through the topological regime [12] [104] [105].

Chapters 5 and 6 address this topic, specifically using molecular  $\text{MoO}_3$  to *p*-dope  $\text{Bi}_2\text{Se}_3$ , demonstrate its stability as a capping layer, and develop a model to describe the charge transfer mechanism.

## 1.7 Current Material Limitations

Recent work about defects and the limitations they place on surface mobility suggest that aggressive chemical mediation is not an avenue to improvement that can be pursued limitlessly. Charged disorder in the  $\text{Bi}_2\text{Se}_3$  system has been studied theoretically [106] and experimentally, both by transport [12] and STM [107]. Kim *et al.* measured mobilities and carrier density through the ambipolar regime ( $n$ - to  $p$ -type conduction transition) and found that the residual carrier density was proportional to the disorder [12]. Because dopant densities in bismuth selenide are typically on the order of  $10^{13} \text{ cm}^{-2}$ , this limits the mobility to  $\sim 1000 \text{ cm}^2/\text{Vs}$ .

## 1.8 Summary & Open Questions

There remains a substantial gap between the bold theoretical predictions for topological insulators and what has been achieved experimentally to date. Substantial progress has been made in understanding the material limitations preventing access to the nontrivial surface states of interest. At the same time, theory pushing past symmetry breaking, further expanding our understanding of topological order, has continued to evolve. Notably, there has been rapidly expanding interest in so called ‘topological crystalline insulators’, labelled as such because lattice symmetries, instead of time reversal symmetry, are the basis of protected metallic surface states [108].

The following chapters of this thesis discuss experiments I performed to un-

derstand exactly how these issues of atmospheric exposure and doping arise and are related to the carrier density of  $\text{Bi}_2\text{Se}_3$  as it is grown in vacuum by molecular beam epitaxy. Additional experiments were performed to study various avenues of reducing the carrier density and passivate samples exposed to ambient conditions. Chapter 2 lays out the novel technique of growing an epitaxial thin film in vacuum while making simultaneous transport measurements, originally appearing as an SPIE conference proceeding [109]. Chapter 3 uses this technique to do a transport based growth study of  $\text{Bi}_2\text{Se}_3$  thin films, to better understand the origins of the persistent  $n$ -doping in thin-film samples. These results appeared in *Applied Physics Letters* [110]. Chapter 4 is a second methods section outlining the technique of X-ray photoemission spectroscopy (XPS), which pertains to the results presented in chapter 5, where XPS was used to study the doping of  $\text{Bi}_2\text{Se}_3$  using the molecular electron acceptor  $\text{MoO}_3$ , subsequently appearing in *ACS Nano* [111]. Chapter 6 discusses the transport aspects of the  $\text{MoO}_3$  study in greater detail. Chapter 7 provides a final summary and outlook of future work.

## Chapter 2: Methods I: Molecular Beam Epitaxy with Simultaneous Electrical Transport Measurements

### 2.1 General Introduction: Molecular Beam Epitaxy

Molecular beam epitaxy (MBE) refers to the process of growing crystalline materials in an ordered fashion (“epitaxially”) on a substrate, through deposition of atomic or molecular reactants in stringent vacuum conditions. This allows for high purity, crystalline, single layer growth. MBE systems are typically equipped with anywhere from two to six reactant sources (or more), effusion cells or electron beam evaporators, a variable temperature growth stage, and apparatus to examine the crystallinity of the substrate and grown material, such as a reflection high energy electron diffraction system (RHEED). The vacuum chamber itself has a liquid nitrogen cooled shroud to maintain pressures in the  $10^{-10}$  Torr or better during the growth process. By varying the relative flux densities of the various sources, the temperature of the substrate, and considering the initial substrate surface quality and lattice constant relative to the crystal structure of the film to be grown, the desired results can be achieved. RHEED gives real time continuous feedback about the crystallinity of the sample; by monitoring the diffraction pattern as it comes in

and out of focus, layer-by-layer control can be achieved.

The experimental challenge with topological insulators is making ready electronic access to the non-trivial edge and surface states, when there is interference from the trivial bulk states as a result of intrinsic material doping, as well as the subsequent effects of ambient exposure. Bismuth selenide exemplifies these challenges, having surface states of interest that are very difficult to separate from the trivial bulk conduction signals, as well as suffering material instabilities making it susceptible to doping when exposed to air, complicating *ex situ* device fabrication.

This project was originally conceived by graduate student Jianhao Chen to develop the ability to grow films by MBE while replacing the typical feedback apparatus (e.g., RHEED) with the ability to make real time, *in situ* electrical contact to the sample during and after growth. The rest of this chapter details how these instrumentation needs were met as well as the changes that had to be made on the sample side to make this possible. The essential feature is the placement of the growth stage simultaneously in a variable magnetic field while being inside the vacuum chamber. This required several key features: a growth stage with an electrical feedthrough for transport measurements, a stencil mask to exactly define the growth region, and the *ex situ* preparation of an insulating growth substrate with pre-wired, electrical contacts. These aspects will be discussed in greater detail in the following sections.

## 2.2 Electrical Transport

The electrical conductivity, carrier type and density, mobility, and how these properties respond as functions of temperature, magnetic field, are the parameter space of transport measurements in the broadest sense. There are a myriad of ways to measure these data; this section focusses on the Hall bar geometry used for the experiments in this thesis. This measurement geometry was chosen because it allows measurement of both components of the 2D resistivity tensor, eliminates the effect of contact resistance in measuring the resistivity tensor, is insensitive to how contact to the sample is made (no strict edge requirement like van der Pauw measurements), and lends itself to the fabrication of a physical stencil mask, the growth through which defines the measurement area.

### 2.2.1 Hall Effect

Considering the Lorentz force is how to understand the transverse Hall voltage and its relationship to the carrier density. An electron (charge  $q$ ) in an electric field  $E$  moving with velocity  $v$  and magnetic field  $B$  feels a Lorentz force  $\mathbf{F}_{\text{Lorentz}} = q(\mathbf{E} + \mathbf{v} \times \mathbf{B})$ .

In the Hall bar geometry, an out of plane magnetic field  $\mathbf{B}$  (figure 2.1, green inset) creates a lateral force ( $\mathbf{F}_{\text{Lorentz}}$ ) on a moving electron ( $\mathbf{v}$ , from the applied current  $I_{SD}$ ). This is balanced by the transverse potential that develops across the sample as a result of the charge accumulation:

$$\mathbf{F}_{\text{Lorentz}} = 0 = q|\mathbf{E}| + q|\mathbf{v}||\mathbf{B}| \rightarrow \frac{qV_{xy}}{W} = -q|\mathbf{v}||\mathbf{B}|$$



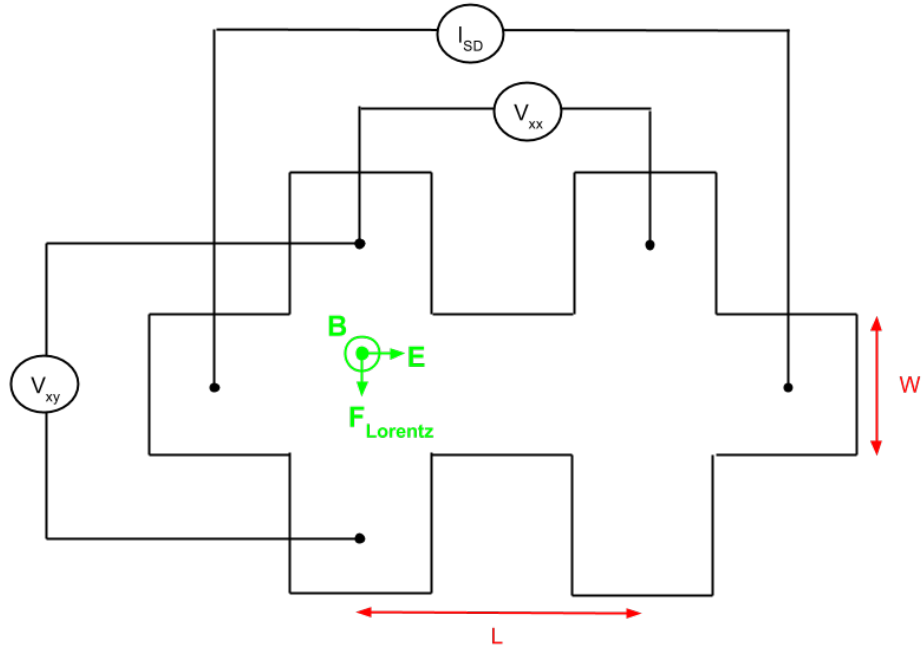


Figure 2.1: Hall bar measurement geometry. The length ( $L$ ) and width ( $W$ ) determine the length/width ratio ( $L/W$ ).  $I_{SD}$  is the source- drain current,  $V_{xx}$  is the longitudinal voltage, and  $V_{xy}$  is the transverse voltage. The (green) inset shows the direction of the electric field, magnetic field (out of plane) and resulting Lorentz force.

Writing the (measurable) current  $I$  in terms of the volume charge density  $n_{3D}$ , geometric factors  $W$ ,  $L$ , and  $t$  (sample thickness), and charge velocity  $|\mathbf{v}|$ :

$$I = qnWt|\mathbf{v}| \rightarrow q|\mathbf{v}| = \frac{I}{n_{3D}Wt}$$

Substituting this in to the Lorentz expression:  $\frac{qV_{xy}}{W} = -\frac{I|\mathbf{B}|}{n_{3D}Wt}$

Solving for  $n$  in the case of electron conduction ( $q = e$ ):

$$\frac{1}{n} = e \frac{dR_{xy}}{dB}$$

Where  $n \equiv n_{3D}t$  is the two dimensional carrier density. Finally, note that in the case of the transverse signal,  $V_{xy} = |\mathbf{E}|W$  and  $I = jW$ , where  $W$  is the width of the Hall bar, so the Hall resistivity is equivalent to the Hall resistance:

$$\rho_{xy} \equiv \frac{|\mathbf{E}|}{j} = \frac{V_{xy}/W}{I/W} = \frac{V_{xy}}{I} \equiv R_{xy} \quad (2.1)$$

So the expression for the carrier density can be written as:

$$\frac{1}{n} = e \frac{d\rho_{xy}}{dB} \quad (2.2)$$

With  $n$ , the mobility can be computed using the sheet resistance data:

$$\frac{1}{\mu} = ne\rho_{xx} \quad (2.3)$$

( $\rho_{xx} = \frac{V_{xx}}{I} \frac{W}{L}$  is the sheet resistivity)

The sign of  $n$  indicates the carrier type (holes or electrons) and its magnitude is a direct measure of the integrated density of states. In the context of  $\text{Bi}_2\text{Se}_3$ , “doping” refers to how far up the Fermi level is in the conduction band, which is

exactly what  $n$  is indicating. Mobility can be interpreted as a figure of merit for the electrical conduction efficiency of a semiconductor which makes it a good measure of disorder. By these measures, “good”  $\text{Bi}_2\text{Se}_3$  samples have low doping ( $n = 0$  when  $E_F = E_D$  in a naive sense) and high mobility.

### 2.2.2 AC vs DC measurements

AC lock-in measurement techniques have the advantage of working under low excitation power, a necessary condition for making measurements at low temperature. In contrast, making measurements at growth temperatures ( $\sim 200^\circ\text{C}$ ) required a completely different approach. Geometric constraints limit how small and well defined the sample can be; the limited magnitude of the magnetic field also conspire to make the Hall signal small and difficult to measure. A  $10\ \mu\text{A}$  positive DC excitation was used in an effort to increase the size of the transverse Hall signal, which still required additional amplification using an SR560 voltage preamplifier. The transverse signal was generated by periodically sweeping the magnetic field through zero to positive and negative maximum values. The macroscopic nature of the sample causes a parasitic contribution of  $\rho_{xx}$  to the  $\rho_{xy}$  signal that was eliminated by subtracting a linear fit to the changing zero-field,  $\rho_{xx}$  signal from the linear fit of the  $\rho_{xy}(B)$  transverse signal.

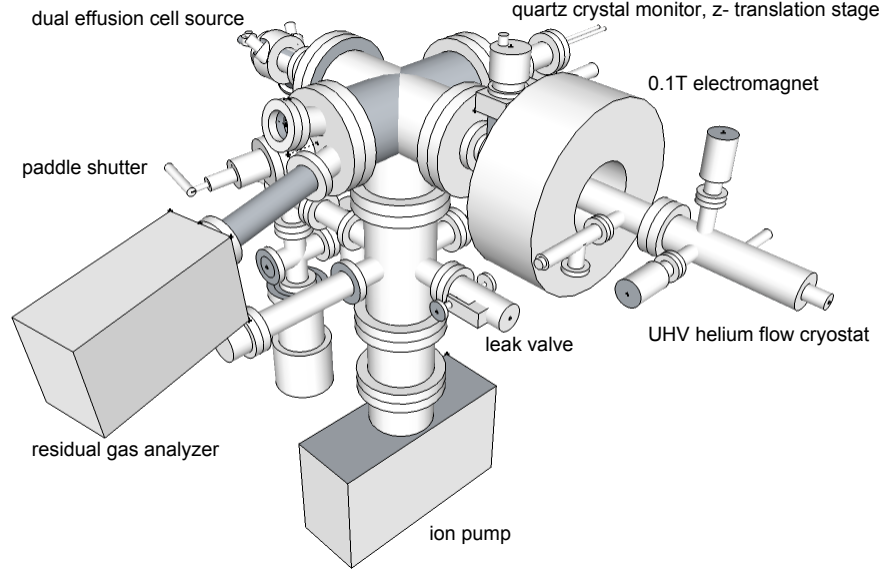


Figure 2.2: Scale diagram of in situ transport thin film growth chamber.

## 2.3 System Specific Details

### 2.3.1 Apparatus

To accomplish this *in situ* transport growth measurement, the apparatus in figure 2.2 was constructed. The working distance between the custom built, dual effusion cell source and the growth substrate is 30 cm. The growth substrate is mounted on a commercial UHV helium flow cryostat (Advanced Research Systems Inc. model no. LT3B-110), sitting centered in the bore of a GMW magnet systems (model no. 11901111) water cooled electromagnet capable of continuously sweeping the magnetic field strength from negative to positive 1100 Gs at the sample. The base pressure is  $5 \times 10^{-9}$  torr; during growth, the chamber pressure will rise to  $1 \times 10^{-7}$

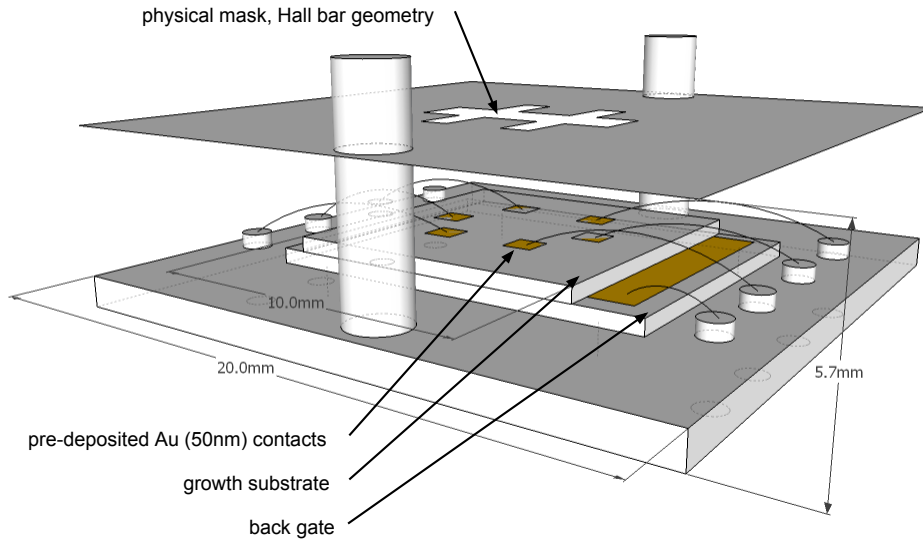


Figure 2.3: Scale diagram of substrate stage for film growth illustrating connections for simultaneous transport measurement.

torr.

### 2.3.2 Growth Stage

Figure 2.3 shows a detail of the growth stage: electrical contacts are pre-deposited on the substrate surface, and wire-bonded to pins connected through the cryostat. A physical mask mounted above the substrate ensures the growth of the film in the Hall bar geometry.

External heat tape at the cryostat, combined with a nichrome wire heater installed internally, directly underneath the growth stage allows for substrate growth temperatures between 100-250°C. Liquid helium cooling can reduce the sample tem-

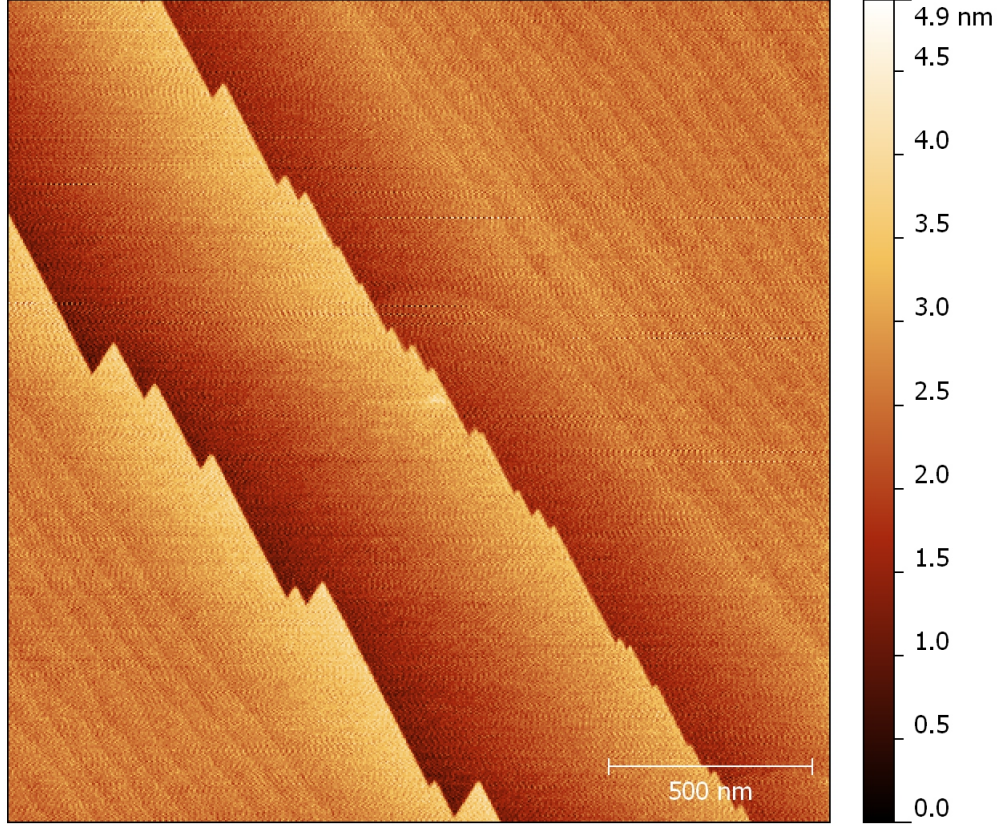


Figure 2.4: Atomic force microscopy (AFM) image of  $\text{SrTiO}_3$  (111) substrate surface after oxygen annealing, showing lattice sized steps.

perature to 15 K. Because of this extreme operating temperature range, a platinum resistance thermometer was installed in the same position as the heater to provide thermal feedback control of the substrate growth temperature.

### 2.3.2.1 Substrate and Contact Preparation

For the experiments described in this thesis,  $\text{SrTiO}_3(111)$  is used as the substrate material. Annealing in oxygen produces an atomically-flat surface suitable for thin film growth [112]. The commercially procured substrate pieces were cleaned in solvent before being placed in a flow-through furnace. Under an oxygen gas flow

rate of 100 sccm, the substrates were heated from room temperature to 1050°C in one hour. 1050°C was maintained for 4-6 hours before cooling back down to room temperature at the natural rate of the furnace. Post annealing atomic force microscopy (AFM) was used to characterize the surface and determine the necessary length of time for the high temperature anneal. Figure 2.4 is a representative AFM image of the results of this annealing procedure, showing the atomically flat surface steps.

After annealing, a (wetting layer) 5 nm Ti (at 0.2 Å/s) and 50 nm Au (2 Å/s) are deposited by electron beam evaporation. A physical mask is used to deposit the contact pads underneath the corresponding ends of the Hall bar stencil mask, as indicated in figure 2.3. This thickness was chosen for ease of wire bonding; surprisingly electrical contact via growth of the sample film onto the gold pads was always reliably established even though the typical film thickness (10 nm) is much less than the gold contact height.

Single crystal strontium titanate has an exceptionally large ( $10^4$ ) dielectric constant at cryogenic temperatures [113], making it possible to capacitively back-gate the film *in situ* despite the 500  $\mu\text{m}$  substrate thickness. The maximum temperature of the cryostat growth stage prevent flash heating of the growth substrate to high temperatures, the typical method used for substrate surface preparation in vacuum [114].

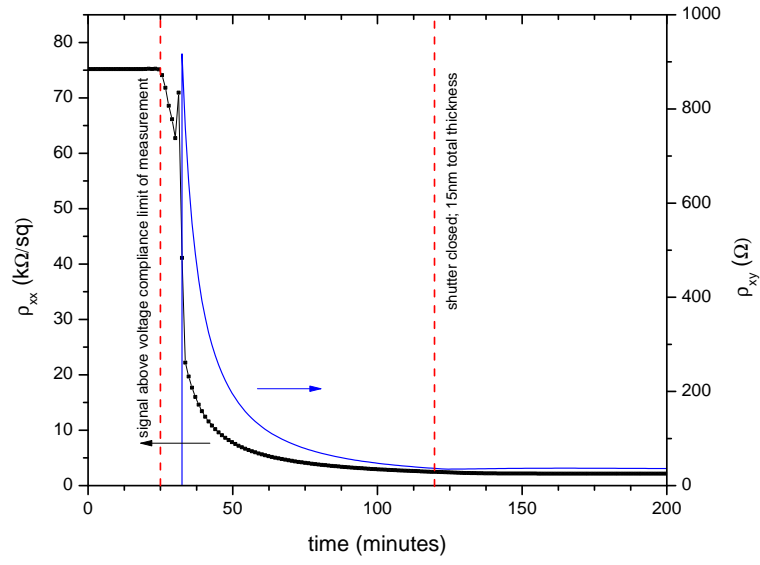


Figure 2.5: Two-dimensional longitudinal (left axis) and Hall resistivity (right axis) measured during film growth.  $10\mu\text{A}$  DC excitation current. The red dashed line at 25 minutes denotes when the sample becomes measurably conductive; before that time, the compliance limit of the measuring instrument ( $>200\text{k}\Omega$ ) is exceeded. The second red line at 120 minutes indicates when the shutter was closed and substrate heating shut off; 15 nm of film was grown.



## 2.4 Demonstration of Technique

Figure 2.5 shows the zero-field, two-dimensional longitudinal ( $\rho_{xx}$ ) and Hall (transverse) resistivity ( $\rho_{xy}$ ) measured as a function of time during film growth. In this instance, the substrate temperature was a constant 115°C throughout the growth, from 0-120 minutes in figures 2.5 and 2.6. At 120 minutes, the shutter was closed and substrate heating shut off; the sample passively cooled to room temperature in three hours. The deposition rates at the substrate surface were nominally 0.1 nm/min for bismuth and 1.8 nm/min for selenium leading to a total film thickness of 12-15 nm. The substrate temperature for film growth is higher than the evaporation point of selenium in vacuum ( $\sim 100^\circ\text{C}$ ) but lower than that of bismuth ( $\sim 600^\circ\text{C}$ ). Bismuth deposition is then the controlling rate of film growth, done in an over-flux of selenium (typically 10-15 times the bismuth rate). Growing more quickly reduces total exposure to impurities/contaminants but compresses the available time for measurements, which are limited by the sweep rate of the magnet. Growing at a rate of minutes per nanometer allows for more data points to be collected during the initial formation stages of the film. The vertical red dotted line at 25 minutes is a guide to the eye to demarcate the time when the film has reliably measurable conductivity: before that time, the resistance between the electrodes is beyond the voltage compliance limit of the measuring instrument ( $>200\text{k}\Omega$ ).  $\rho_{xy}$  is measured while periodically sweeping the magnetic field: the slowly changing background due to parasitic  $\rho_{xx}$  contribution is subtracted out before fitting the  $\rho_{xy}(B)$  data to extract the Hall carrier density.

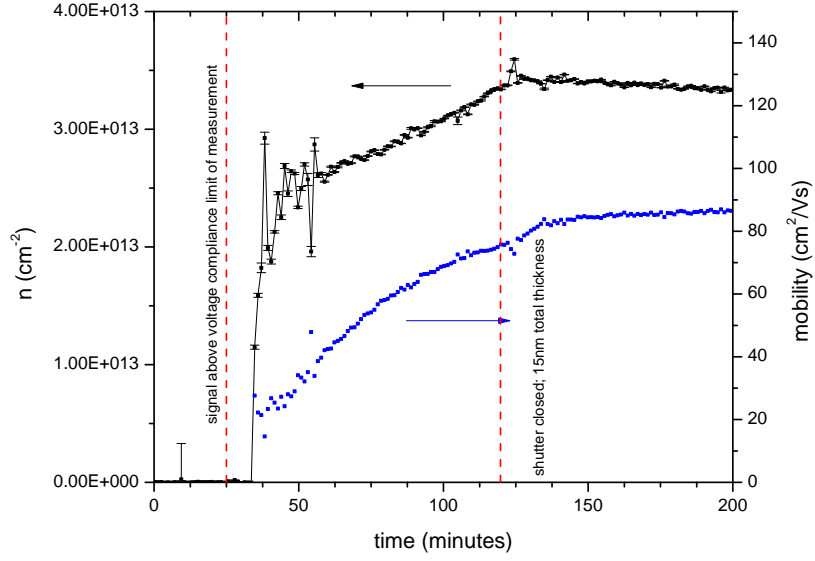


Figure 2.6: Two dimensional (electron) Hall carrier density (left axis) and mobility (right axis) plotted as a function of film growth time. Red dashed lines the same as in figure 2.5.

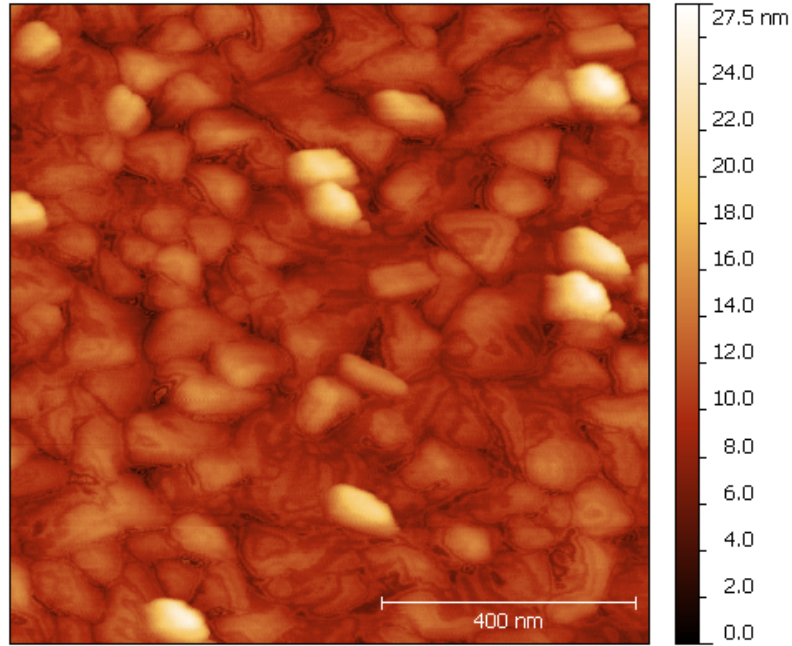


Figure 2.7: ex situ contact mode AFM image of  $\text{Bi}_2\text{Se}_3$  film on  $\text{STO}(111)$

$\mu$  and  $n$  are plotted as a function of film growth time in figure 2.6. For the growth in figures 2.5 and 2.6 we observe that the carrier density has a sharp increase to  $\sim 2 \times 10^{13} \text{ cm}^{-2}$  at the onset of conduction, followed by a slower increase of  $1.2 \times 10^{12} \text{ cm}^{-2}$  for each nanometer of additional film growth. Figure 2.7 is an *ex situ* ambient AFM image of one of the first  $\text{Bi}_2\text{Se}_3$  films successfully grown on strontium titanate STO(111) by this technique. These initial observations are studied in greater detail in chapter 3.

## 2.5 Source-Swapping for Capping Layer Deposition

The geometry of this system allows for vacuum to be maintained in the sample space while the source effusion cell is swapped. This aspect is exploited in chapter 6: after the growth and characterization of a 15 nm  $\text{Bi}_2\text{Se}_3$  sample, the effusion cell was changed to deposit a molecular  $\text{MoO}_3$  overlayer. In light of the problems in this material system with respect to ambient exposure, this methodology provided an unambiguous determination of the efficacy of  $\text{MoO}_3$  as a *p*-doping capping layer.

## Chapter 3: Results I: *In Situ* Bi<sub>2</sub>Se<sub>3</sub> Growth Dynamics Study

### 3.1 Outline of Parameter Space Studied

Using the apparatus previously described in chapter ??, we studied the real time evolution of carrier density and mobility as a function of film thickness during growth for different substrate temperature profiles. The dependence of transport parameters on film thickness have been reported previously for Bi<sub>2</sub>Se<sub>3</sub> on Si(111) and Al<sub>2</sub>O<sub>3</sub> substrates [72] [115], however in those studies multiple films of difference thicknesses were measured *ex situ*, and only one growth parameter regimen could be explored.

Samples were grown using a two-temperature method [116] [117] [118] where the first two nanometers are grown at 110°C and the remainder of the film grown at different temperatures between 200-230°C. I also measured a single-temperature growth at 210°C for comparison. The initial carrier density at the onset of conduction is approximately  $1.5 \times 10^{13} \text{ cm}^{-2}$ , and the carrier density growth with thickness corresponds to a bulk dopant density of  $\sim 5 \times 10^{11} \text{ cm}^{-2}$  per quintuple-layer unit (QL), with both values roughly independent of growth conditions. This suggests that a significant component of the observed doping in Bi<sub>2</sub>Se<sub>3</sub> films is associated with the formation of the Bi<sub>2</sub>Se<sub>3</sub>-substrate and Bi<sub>2</sub>Se<sub>3</sub>-vacuum interfaces, even without expo-

sure to ambient. In contrast, the mobility depends strongly on growth conditions, with maximum mobility achieved for a two-step growth at 110°C and 210°C, consistent with the optimum growth temperatures observed by reflection high energy electron diffraction (RHEED) aided growth studies. [119] [116] [117] [72] [115] [118] The results are strongly suggestive that the charge carrier mobility in our Bi<sub>2</sub>Se<sub>3</sub> films (and likely in the films prepared in other laboratories) is not limited by ionized impurities (dopants) but rather by structural uniformity. Atomic force microscopy (AFM) confirms a strong correlation between mobility and structural coherence of the films.

### 3.2 *In Situ* Transport for Different Growth Parameters

Two-stage growth is carried out at a temperature of 110°C for the first two nanometers of film deposition, before increasing the substrate temperature to a series of ‘hot’ temperatures between 200 and 230°C. Two additional samples were grown: in one instance the substrate temperature was maintained at 110°C, for the other, the substrate temperature was 210°C the entire time, forgoing the initial low temperature growth. Longitudinal and Hall resistivities are monitored continuously during film growth by periodic application of a magnetic field swept continuously between  $\pm 1100$  Gauss. The thickness of each sample is calculated from the bismuth rate using the empirically determined tooling factor for the growth chamber.

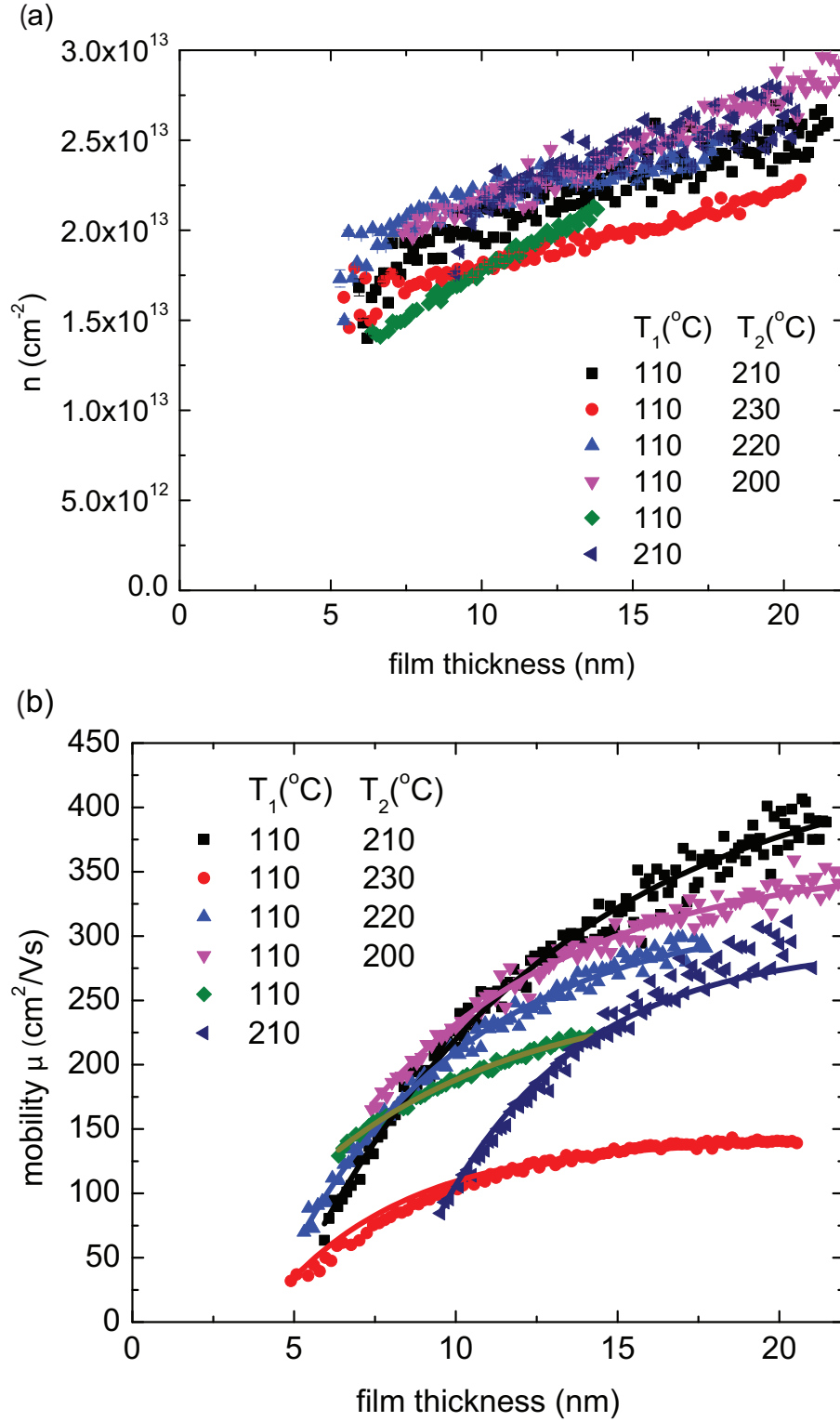


Figure 3.1: (a) Hall carrier density  $n$  and (b) mobility  $\mu$  versus Bi<sub>2</sub>Se<sub>3</sub> film thickness for various growth temperature profiles as shown in legend:  $T_1$  is the substrate temperature for the first 2nm of film deposited,  $T_2$  is the temperature for the remainder of the growth. In the cases where there is no  $T_2$  the  $T_1$  temperature was maintained for the whole growth.

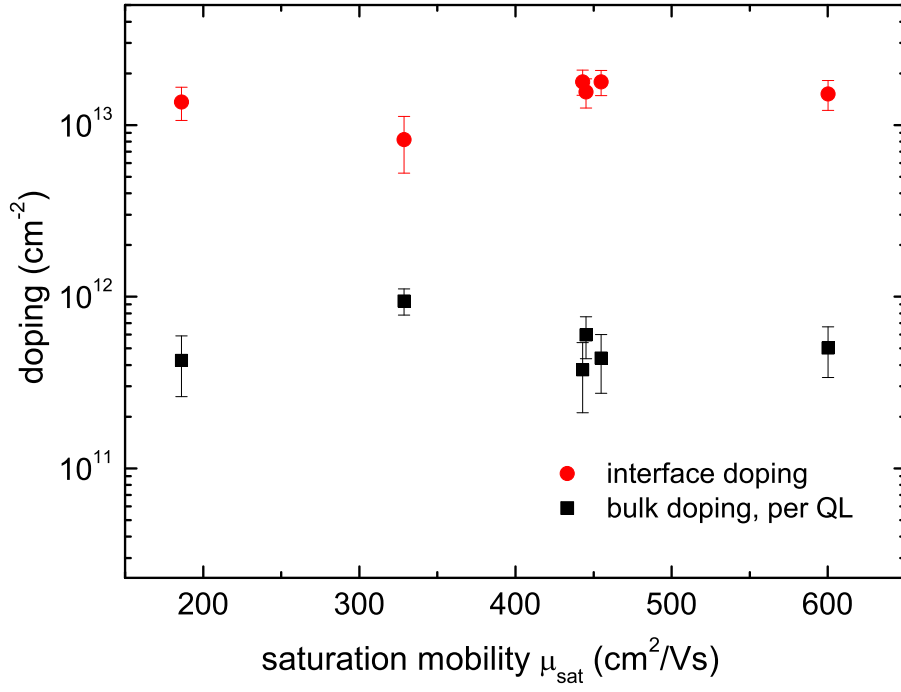


Figure 3.2: Bulk dopant concentration per QL  $dn/dt$  (black squares) and interfacial dopant concentration  $n_0$  (red circles) extracted from the slope and intercept, respectively, of linear fits to  $n(t)$  in fig. 3.1(a), plotted against saturation mobilities extracted from the data in fig. 3.1(b).

### 3.3 Interfacial vs. Bulk Doping

Figure 3.1(a) shows  $n(t)$  and fig. 3.1(b) shows  $\mu(t)$  for the different growth temperature profiles (there are large fluctuations in the magnitude of  $\rho_{xy}$  with time during the initial growth when  $\rho_{xx} > 100$  k $\Omega$ , typically for the first 5 nm of film growth, where presumably the film is highly non-uniform in thickness or possibly discontinuous. The reported data is for when  $\rho_{xx} < 100$  k $\Omega$ , and when  $\rho_{xy}$  values become stable and presumably the film is continuous and uniform). As shown in fig. 3.1(a),  $n(t)$  is very similar for all growth temperature profiles studied.  $n(t)$  is roughly linear, with a positive intercept at zero thickness. The results are in contrast to *ex situ* measurements of Bi<sub>2</sub>Se<sub>3</sub> on Si(111) where  $n \sim \sqrt{t}$  [115], and Bi<sub>2</sub>Se<sub>3</sub> on Al<sub>2</sub>O<sub>3</sub> where  $n$  is independent of thickness [72]. A linear fit to the data in fig. 3.1(a) can be interpreted by ascribing the intercept  $n_0 \approx (1-2) \times 10^{13}$  cm<sup>-2</sup> as an interface dopant concentration which is present as soon as the film is formed, and the slope  $dn/dt \approx 5 \times 10^{11}$  cm<sup>-2</sup> per QL as a bulk (3D) dopant concentration which scales linearly with thickness indicating a constant rate of new dopants added as new film is deposited. It is not possible to deduce from these measurements whether the interface dopants correspond to the Bi<sub>2</sub>Se<sub>3</sub>-vacuum or substrate-Bi<sub>2</sub>Se<sub>3</sub> interfaces, or whether the interface dopants are at the interface itself or incorporated in the first few nanometers of film which may be more disordered. Regardless of the source, the *in situ* nature of our measurements shows that the doping is inherent to the initial formation of the interfaces and not due to reaction with ambient.



### 3.4 Dependence of Mobility on Growth Parameters

Figure 3.1(b) shows the thickness dependence of the mobility  $\mu(t)$ . In contrast to  $n(t)$ ,  $\mu(t)$  shows appreciable changes with different growth temperature profiles. Note that the reported mobilities are measured at the respective growth temperatures, explaining their somewhat lower magnitude relative to mobilities reported elsewhere in the literature. Solid lines in fig. 3.1(b) are fits to the measurement using the phenomenological relation [115]:

$$\mu(t) = \frac{\mu_{sat}}{1 + \frac{\lambda^*}{t-t_0}} \quad (3.1)$$

where  $\mu_{sat}$  is interpreted as the saturation mobility in the thick-film limit. The uncertainties expressed in figures 3.2 and 3.4 for the interfacial and bulk doping, saturation mobility values respectively were estimated from the variance of a series of samples grown with nominally identical (110-210 °C) growth conditions.  $\mu_{sat}$  varies by a factor of 3 for different growth conditions, while the carrier concentration at a given thickness varies by less than a factor of 1.5. This result is surprising, since the carrier concentration is expected to reflect the concentration of ionized impurities (dopants), which have been suspected as a major source of disorder (i.e. lowered mobility) in  $\text{Bi}_2\text{Se}_3$  [12]. These results show that mobility varies strongly with growth conditions though dopant concentrations are nearly constant.

### 3.5 Structural Coherence Studied by Atomic Force Microscopy

To further explore the correlation between dopant concentration and mobility, fig. 3.2 shows the bulk dopant concentration  $dn/dt$  and interfacial dopant concentration  $n_0$  extracted from fig. 3.1(a) as a function of the  $\mu_{sat}$  extracted from fig. 3.1(b). There is no clear correlation of either  $dn/dt$  or  $n_0$  with  $\mu_{sat}$ , in fact the highest and lowest-mobility films have very similar concentrations of both bulk and interfacial dopants. This suggests that dopants are not the dominant disorder in these films, and that the changes in mobility with growth conditions are controlled by another variable.

Atomic force microscopy (AFM) was used to further investigate the crystal quality of the films as another possible variable controlling mobility. Figure 3.3 shows representative *ex situ* AFM images taken using a Bruker Dimension Icon using the contact mode of imaging. A clear difference is immediately evident between the two images: the sample grown at 110-210°C for which mobility is optimal shows evidence of van der Waals epitaxial growth, with atomically flat terraces of width  $\sim 50$  nm, while the sample grown at 110-230°C with lower mobility shows greater roughness and evidence of some crystallites with c-axes oriented non-parallel to the film normal. The differences are quantified using the height-height correlation function  $H(x)$  (function of point-to-point distance  $x$ ) computed for each sample and fit to:

$$f(x) = 2\sigma^2(1 - e^{-(x/T)^2}) \quad (3.2)$$

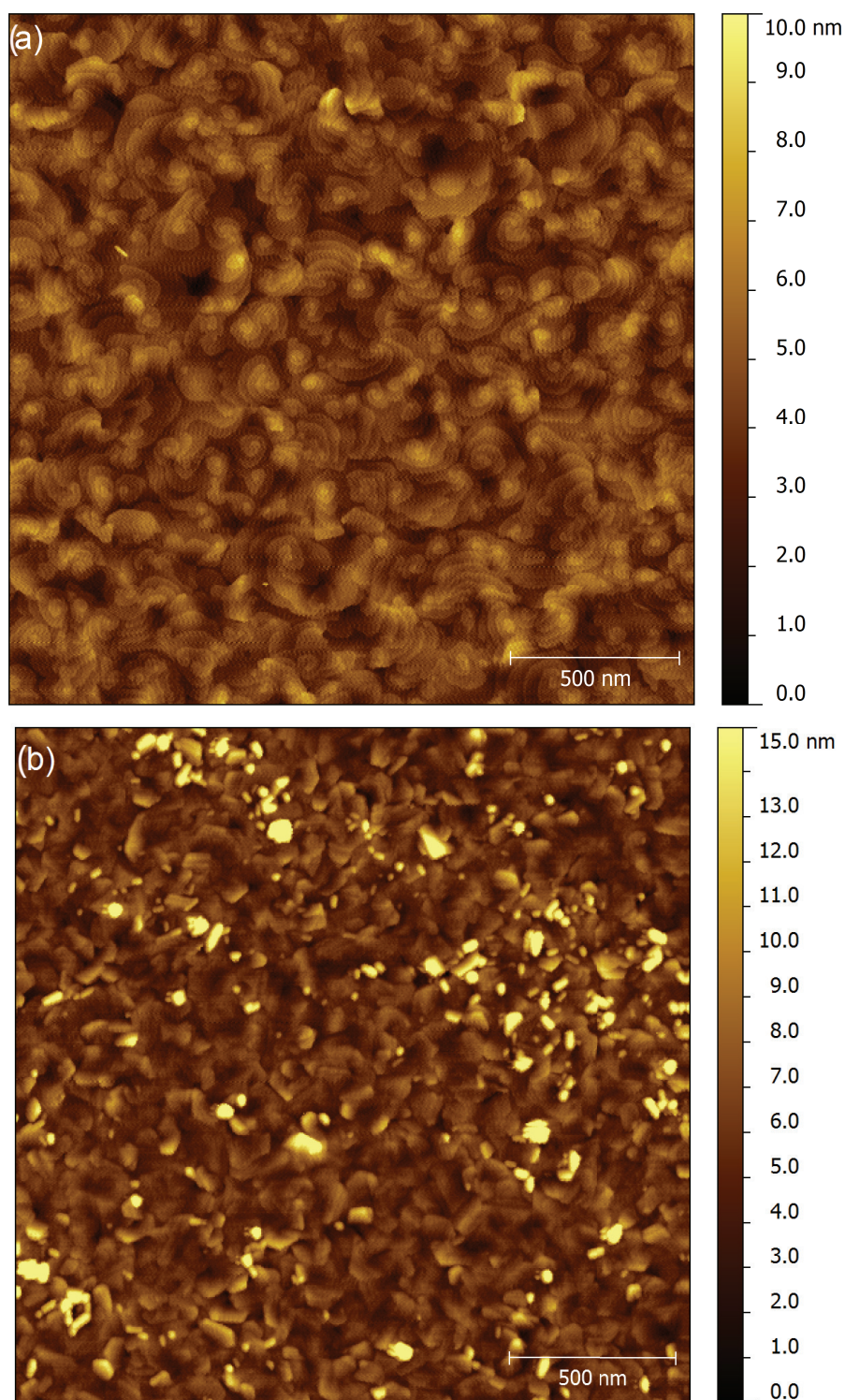


Figure 3.3: *ex situ*  $2\mu\text{m} \times 2\mu\text{m}$  atomic force microscope images of the a) 110-210°C and b) 110-230°C films.

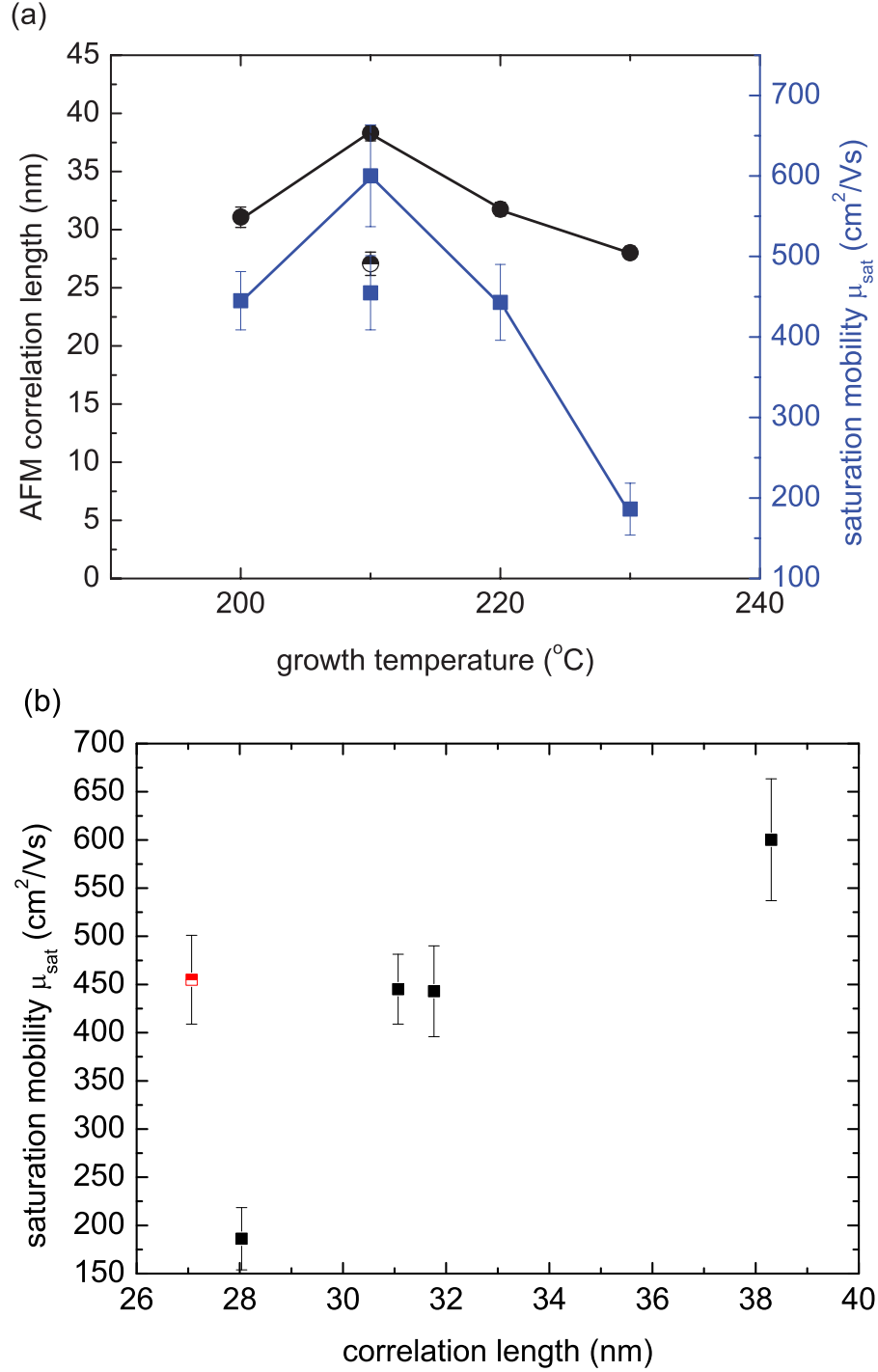


Figure 3.4: (a) Saturation mobilities  $\mu_{sat}$  (squares) extracted from the data in fig. 3.1(b) and correlation lengths  $T$  (circles) from AFM images are plotted as a function of the highest temperature during growth for each respective sample. The half filled points at 210 $^{\circ}\text{C}$  correspond to the single-step growth sample. (b) Saturation mobility plotted against AFM correlation length for each sample; black points are two-step growths, the red half-filled point is the single-step growth at 210 $^{\circ}\text{C}$ .

In this case  $\sigma$  is the standard deviation in height, and  $T$  is the surface correlation length.  $T$  is interpreted as a measure of the structural coherence of the film. For an epitaxial film,  $T$  would be proportional to the mean terrace width.

Figure 3.4(a) shows the AFM correlation length and saturation mobility  $\mu_{sat}$  as a function of the hot growth temperature for each sample, showing a peak both in the mobility and the correlation length for the two-step growth at 110-210°C. The single-step 210°C growth values are also shown, measurably lower than the two-step growth terminating at the same temperature. To further explore this relationship, in fig. 3.4(b)  $\mu_{sat}$  is plotted against the AFM correlation length  $T$ . A high degree of correlation between  $\mu_{sat}$  and  $T$  is observed for the two-step samples, strongly indicating that the structural coherence of the film is the driving variable behind mobility variation with growth conditions.

### 3.6 Summary of Findings

Bismuth selenide films grown according to widely employed methodology within the literature are substantially  $n$ -doped as grown in vacuum, before any exposure to atmosphere. Significant interface ( $\sim(1-2) \times 10^{13} \text{ cm}^{-2}$ ) and bulk ( $\sim 5 \times 10^{11} \text{ cm}^{-2}$  per QL) dopant concentrations are observed, nearly independent of growth conditions. Differences in mobility with respect to growth temperature appear to be driven by changes in film morphology, as quantified by *ex situ* AFM analysis, rather than variations in dopant concentrations. The implication is that work to date on  $\text{Bi}_2\text{Se}_3$  film preparation has optimized the structural coherence of the films,

but significant gains remain to be made in reducing both bulk and interfacial dopant concentrations.

## Chapter 4: Methods II: Photoemission Spectroscopy (XPS)

### 4.1 Introduction

This chapter presents the relevant background theory and methods for synchrotron based X-ray photoelectron spectroscopy (XPS), in the context of the results presented in chapter 5, namely, manipulating the doping of single crystal bismuth selenide with an overlayer of molecular  $\text{MoO}_3$ . These experiments were done at the soft x-ray end station of the Australian Synchrotron.

The photoelectric effect is the phenomena of electrons being emitted from conducting materials when they are exposed to photons. In an XPS experiment, samples are exposed to a light source, and the energy of the photoemitted electrons are measured (see figure 4.1). These data are analyzed to infer information about the nature of the electronic structure of the sample. In this specific instance XPS was used to measure Bi, Se, and Mo core level binding energies (see figure 4.3 as a representative example of core level spectra) and secondary electron cutoff, quantities that provide information about the doping and work function of  $\text{Bi}_2\text{Se}_3$  and how these changed when  $\text{MoO}_3$  was introduced, as well as the fraction of  $\text{MoO}_3$  molecules that participated in the doping process.

The subsequent sections of this chapter detail a little bit more about the

photoemission process, how these samples are prepared and the data is collected, as well as how the measured spectroscopy peaks are analyzed.

## 4.2 Modeling the Photoemission Process

Figure 4.1 illustrates the basic quantities measured in an XPS experiment. The synchrotron provides high intensity incident photons at a selectable energy  $h\nu$ . The detector measures the kinetic energy of the photoemitted electrons. This zeroth order picture of XPS is too simple for meaningful use of the technique for the present work.

Figure 4.2 illustrates an incrementally more sophisticated way of thinking about this process. Photoexcitation is described using Fermi's Golden Rule; treating the penetration through the surface into the free electron state separately makes the initial problem more analytically tractable.

From a user standpoint this is a series of solved problems. Both the incident light and the detector (SPECS) have been extensively characterized. The synchrotron has selectable incident photon energy (in contrast to a fixed excitation from the lab based source). When changes to the photon energy are made it is standard practice to confirm the selected photon energy on a gold standard by measuring the Au 4f<sub>7/2</sub> core level peak (84.00 eV), or Fermi edge (figure 4.3). This allows the binding energy scale of all Bi<sub>2</sub>Se<sub>3</sub> related spectra to be referenced to the Fermi energy ( $E_F$ ) by having the Au reference in electrical contact with the sample. For most photon energies the Au 4f peak is the preferred reference standard. At



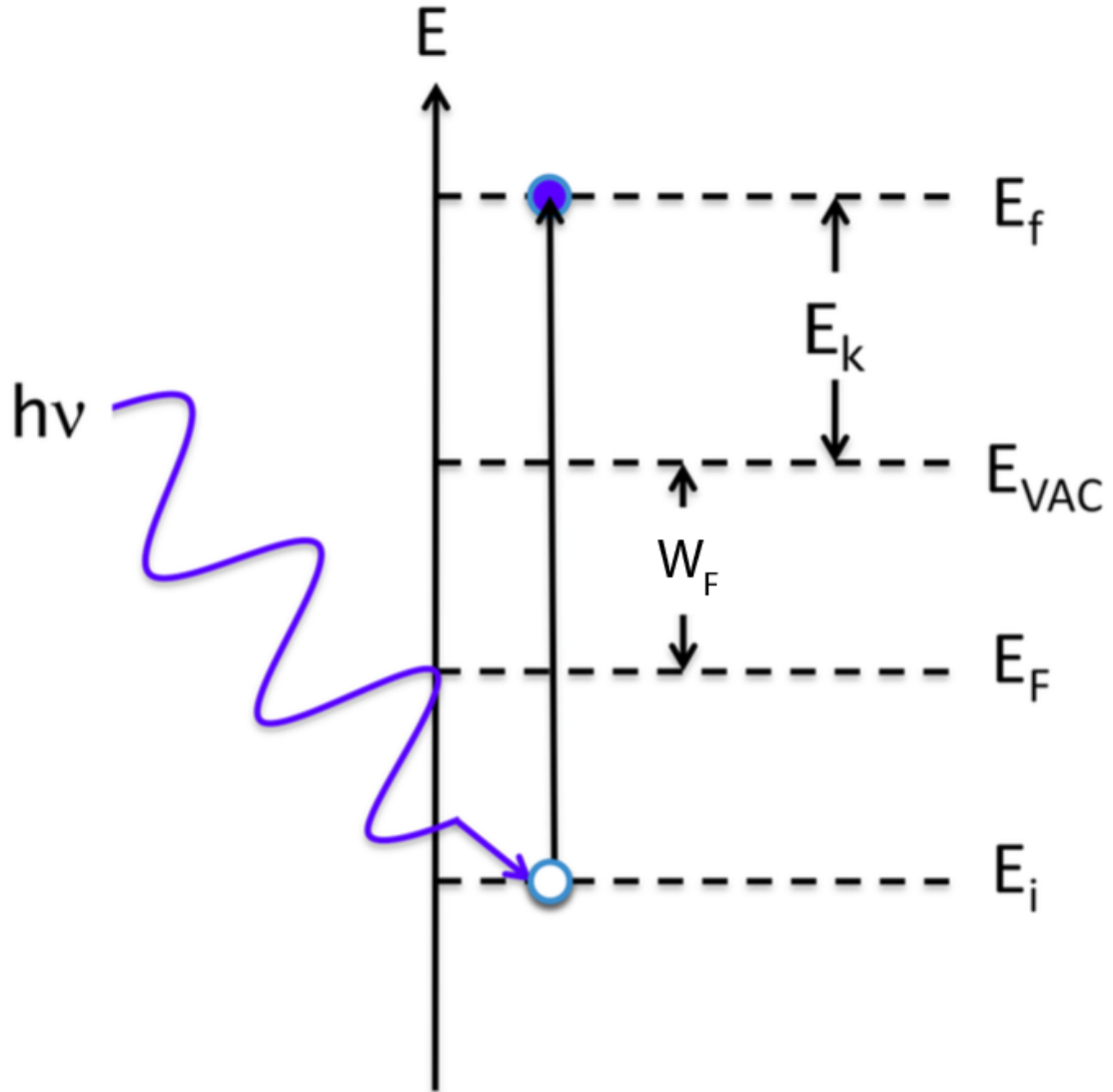


Figure 4.1: Illustration of the photoemission process for an incident photon of energy  $h\nu$ . The photoemitted electron has initial energy  $E_i$  and final energy  $E_f$ . The work function  $W_F = E_{vac} - E_F$  is the energy difference between the vacuum energy ( $E_{vac}$ ) and Fermi level ( $E_F$ ). The kinetic energy of the electron  $E_k = E_f - E_{vac}$  is the difference between the final energy ( $E_f$ ) and vacuum energy. Taken from ref. [15].

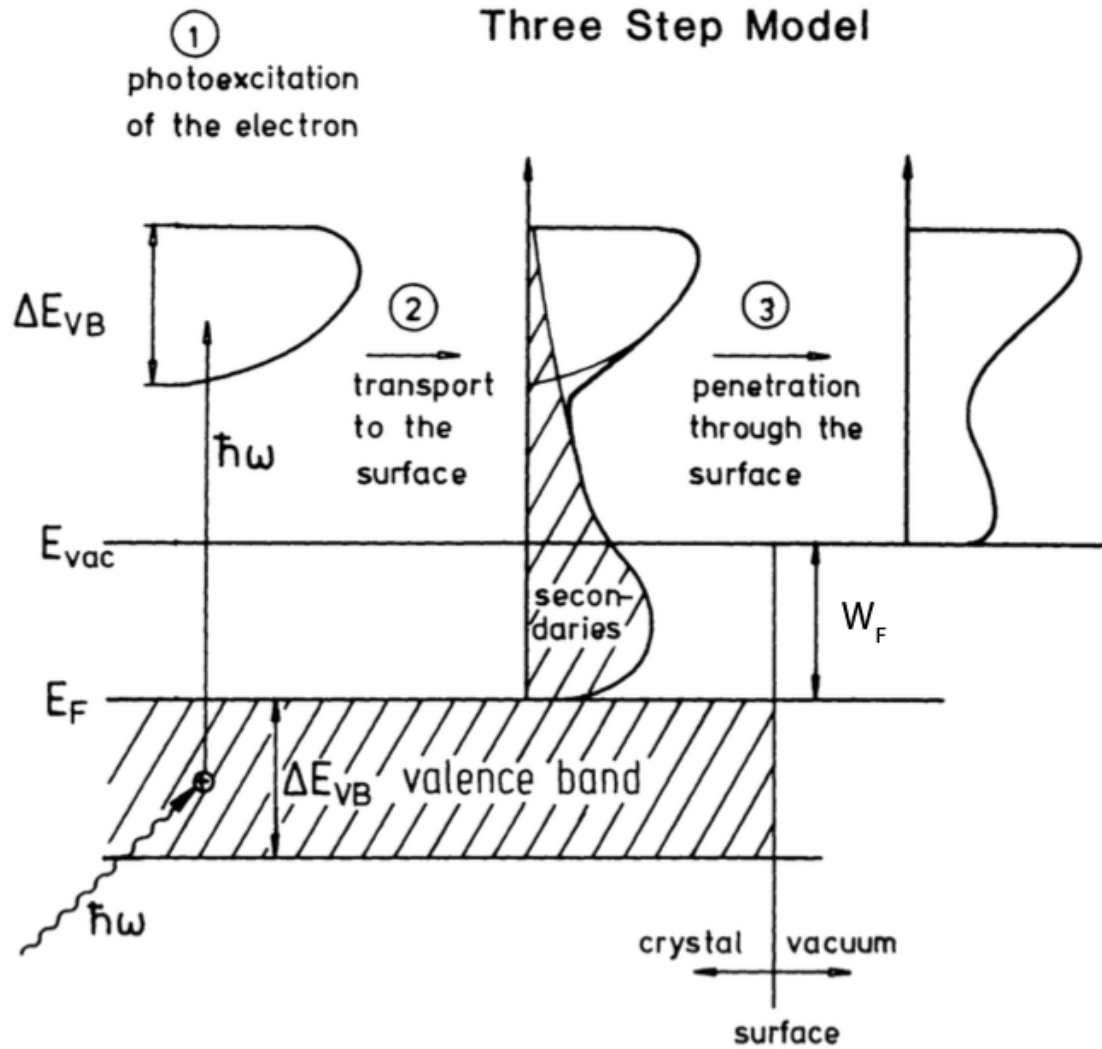


Figure 4.2: The “Three Step Model” divides the photoemission process into three separate analytical problems. From left to right, in chronological order: 1) photoexcitation of the electron, 2) transport of the electron to the surface, and 3) penetration through the surface into free vacuum. Reproduced from ref. [16].

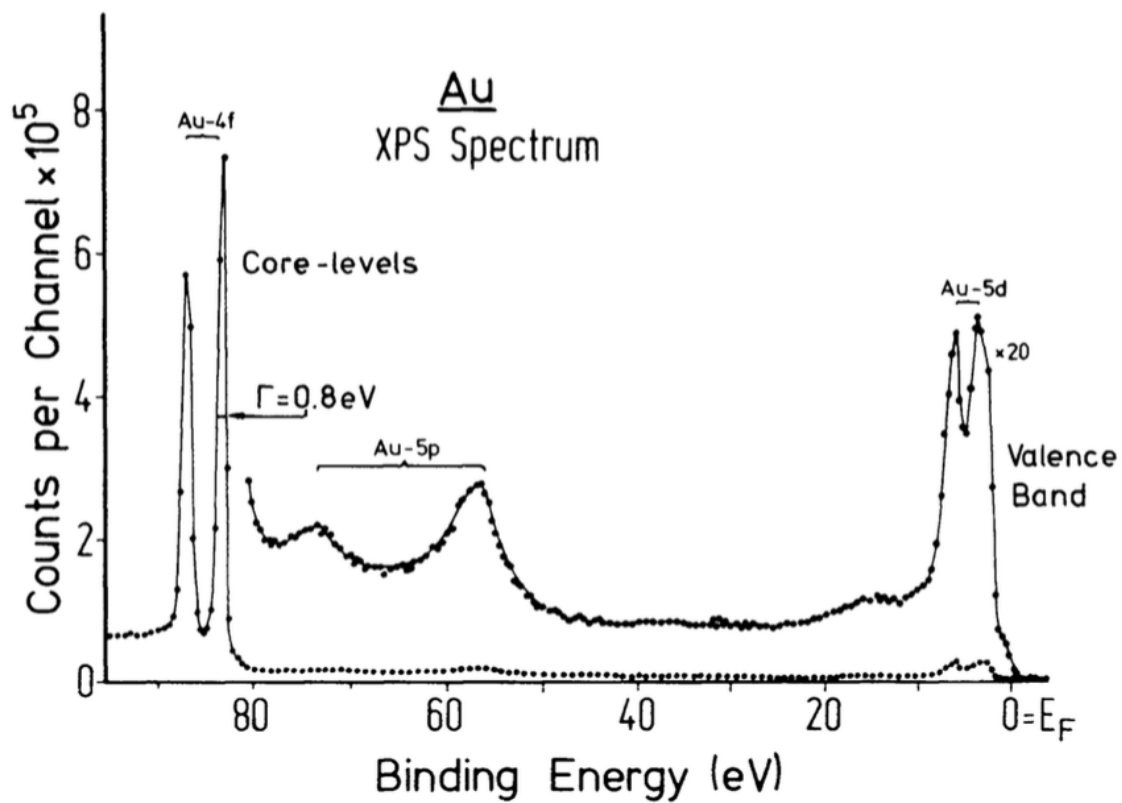


Figure 4.3: Example of XPS spectroscopy for polycrystalline Au sample. Reproduced from ref. [16].

lower photon energies, for example the 100 eV used for some of these experiments, the 4f binding energy of 84 eV is too close to be safely measured by the detector. The threshold would be below the secondary electron cutoff, meaning the detector would be flooded with excess electron signal, damaging the detector and distorting the desired measurement. Instead, the Fermi edge (through zero binding energy) is measured and fit to the Fermi-Dirac distribution. Prior to performing experiments a detailed beam damage study was performed to ensure the molecules were stable under prolonged beam exposure.

### 4.3 Surface Sensitivity, Auger Peaks, Photon Energy Selection

This study was performed at the synchrotron (as opposed to using a lab based instrument) because surface sensitivity was an important factor in performing a surface doping study. The continuously tunable photon energy available at the synchrotron enables surface sensitivity to be optimized as follows.

The typical energy range for XPS work is between 10 and 2000 eV (the soft X-ray bandwidth). Figure 4.4 makes clear why this technique is acutely surface sensitive: this so-called “universal curve” of mean free path for electrons can be as little as a few angstroms for the right excitation energy. A consequence of this surface sensitivity is the need for ultrahigh vacuum (UHV) conditions during measurement; the presence of surface contamination will further disrupt the unscattered primary photocurrent, reducing or even destroying the signal of interest. The UHV environment was an advantage in this instance because of the ambient instability of

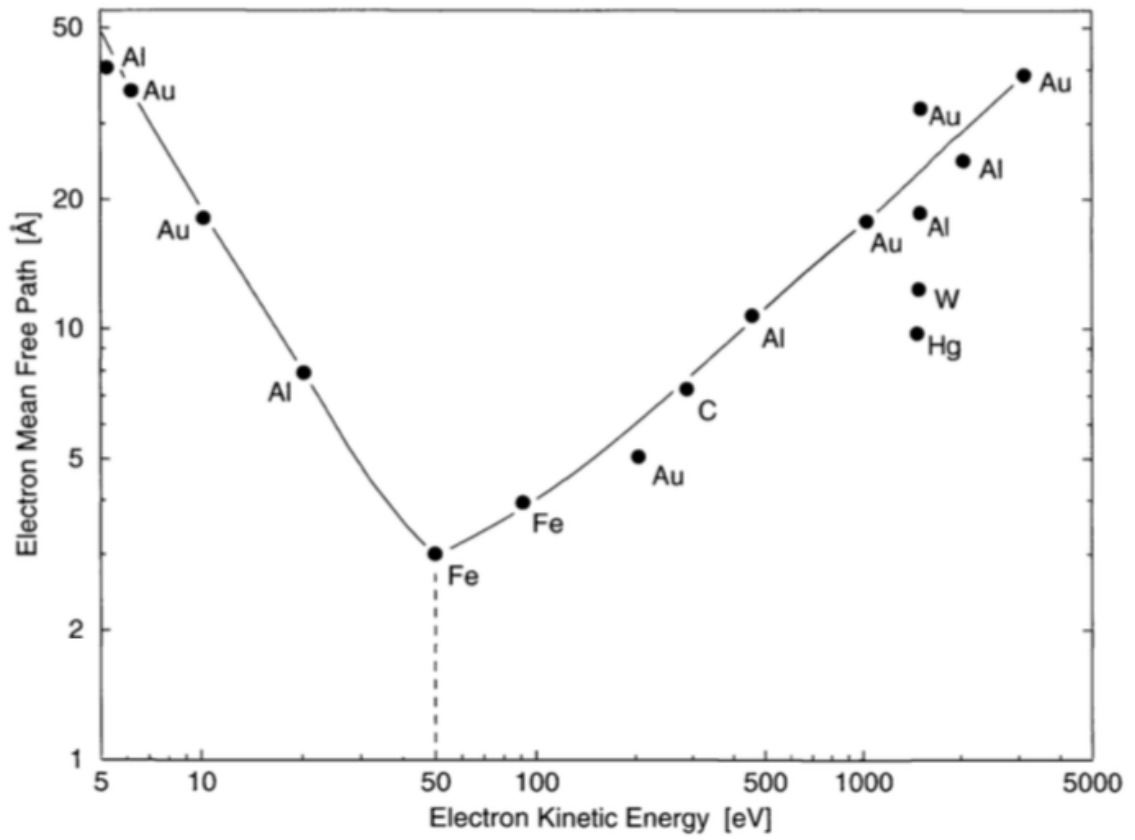


Figure 4.4: Electron mean free path plotted against photon energy for different metals. Taken from reference [16].

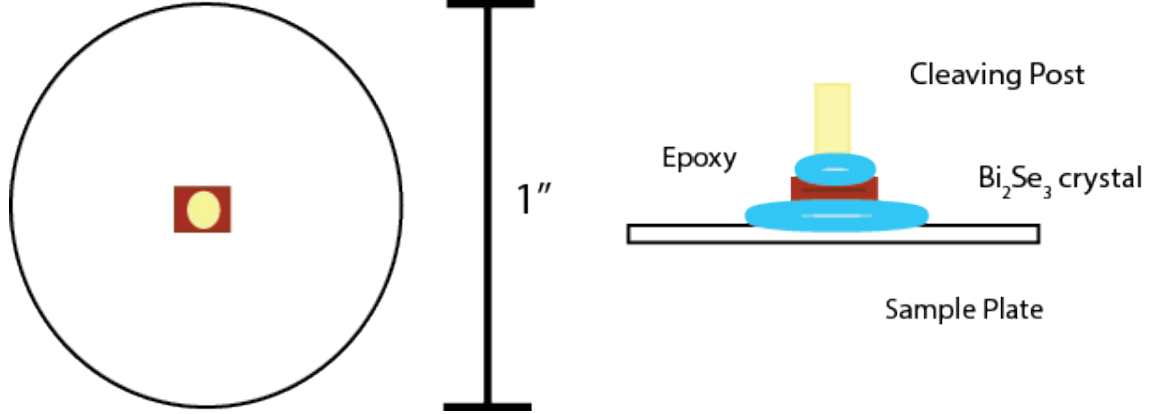


Figure 4.5: Schematic of single crystal  $\text{Bi}_2\text{Se}_3$  flake (red) mounted on a synchrotron sample plate for *in situ* cleaving. Care is taken to avoid epoxy for the post covering the edges of the sample flake to ensure a clean cleave requiring relatively little force.

bismuth selenide.

Different elements have different characteristic core level spectra, one of the central utilities of this technique. In addition to the primary photoemission process, there are various inter- and intrastate transitions, in this context referred to as Auger electron spectra. These Auger peaks can coincide with core levels of interest. This complicates the selection of photon energy beyond merely the penetration depth/surface sensitivity: energies have to be selected so Auger peaks don't collide with the core levels of interest.

#### 4.4 Single Crystal Sample Preparation

These  $\text{Bi}_2\text{Se}_3$  crystals were obtained from Paul Syers and Prof. Johnpierre Paglione of the University of Maryland. They were low-doped (carrier density  $\sim 10^{17} \text{ cm}^{-3}$ ) with bulk resistivity exceeding  $2 \text{ m}\Omega \text{ cm}^{-1}$  at 300 K were grown by melting high purity bismuth (6N) and selenium (5N) in sealed quartz ampoules [120].

Figure 4.5 illustrates how  $\sim 3 \times 3$  mm flakes were mounted to metal plate sample holders using a vacuum compatible conductive epoxy (Epotek H22). The flakes are mounted according to the preferred cleaving plane of the bismuth selenide to ensure a flat, high quality surface for the subsequent measurements. 4mm diameter stainless steel rod was cut into  $\sim 1$  cm lengths and filed at the end for a uniform bonding surface.

The samples were then introduced into the ultra-high vacuum photoemission endstation where they were subsequently cleaved *in situ* by using a mechanical manipulator to knock off the stainless steel cleaving post and kept at pressures  $< 10^{-9}$  mbar during experiments and deposition.

F4-TCNQ and  $\text{MoO}_3$  (Sigma Aldrich) were deposited onto the cleaved samples at room temperature using commercial effusion cells (MBE Komponenten GmbH) operating at 135 °C and 525 °C respectively, with a quartz crystal microbalance (QCM) used as a guide for the deposition rate.

## 4.5 Secondary Electron Cutoff to Determine Work Function

In figure 4.2 one can think of the work function  $W_F$  as a threshold that eliminates a large portion of the secondary electrons produced in the photo excitation process in moving from step two to three. This secondary electron cutoff threshold can be measured to determine the work function using the following relation:

$$E_{kin} = \hbar\omega - W_{\text{Bi}_2\text{Se}_3} - |E_B|$$

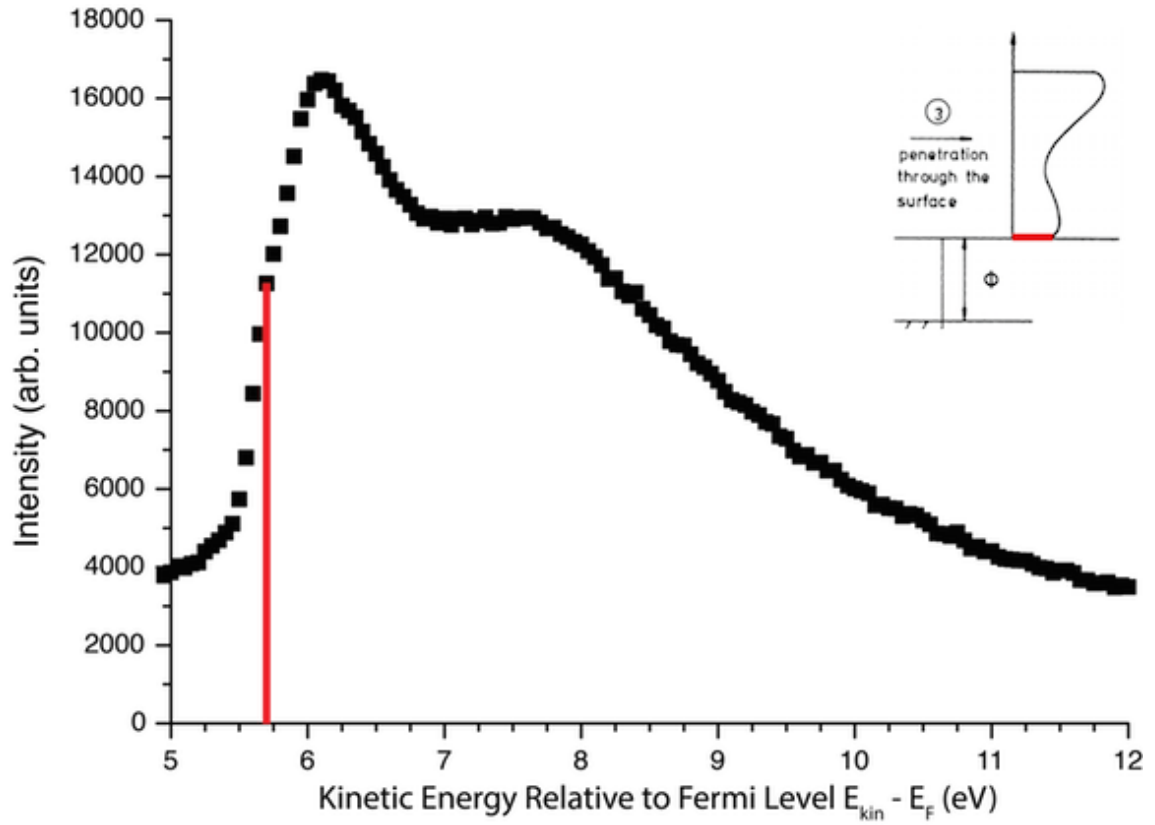


Figure 4.6: Example of secondary electron cutoff spectra taken on cleaved single crystal  $\text{Bi}_2\text{Se}_3$  (100 eV photon energy). Compare for example to figure 1.7 in reference [16]. Inset is “step 3” from figure 4.2 with the red line highlighting the work function energy being measured.



$\hbar\omega$  is the (known) photon energy and the binding energy  $E_B$  is zero at the Fermi level. The cutoff is named for the threshold at which the kinetic energy  $E_{kin}$  is zero (at the vacuum energy  $E_{vac}$ ), therefore determining the work function  $W_F$ . This is demonstrated in figure 4.6, a secondary electron cutoff measurement performed on  $\text{Bi}_2\text{Se}_3$ . The cutoff energy is indicated in red.

The SPECS detector at the soft X-ray end station where this work was done has an intrinsic work function of 4.5 eV. Because this is quite close in energy to the bismuth selenide work function, the sample is biased (-8V in this case) to experimentally deconvolve the two quantities. The experimental uncertainty is  $\pm 0.03$  eV for this measurement.

## 4.6 Peak Fitting

XPS data is peak fit in many different ways depending on the phenomenological circumstances. In this case, the core level positions and total areal intensity give information about doping changes and thickness of the molecular overlayer. The secondary electron emission contributions are a background that is removed before fitting the remaining peaks to a convolution of Lorentzian and Gaussian functions called a Voigt function [121]. Practically speaking this can be accomplished by a handful of different commercially available analysis programs. The data in this work was analyzed with Origin peak fitting routines.

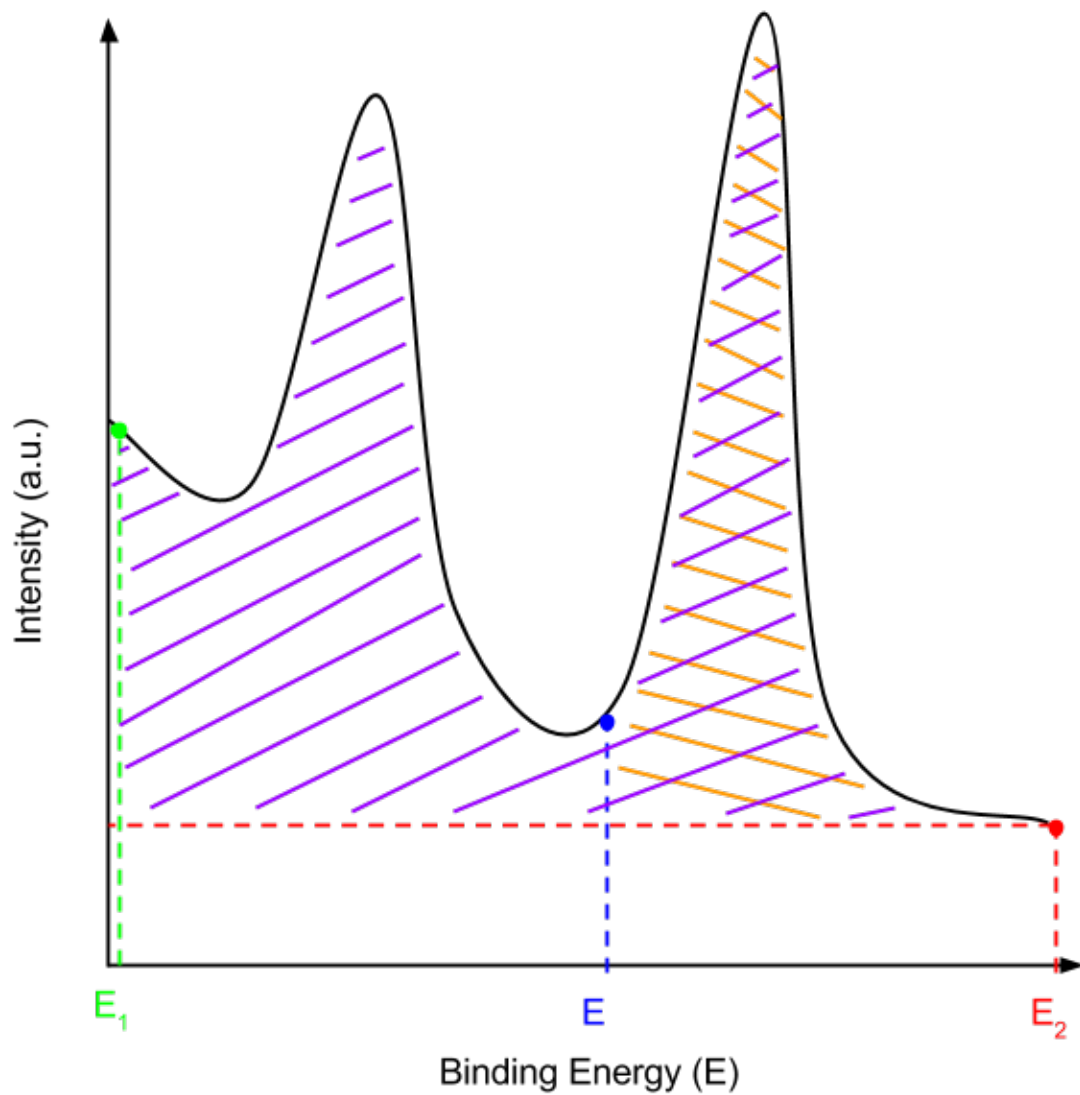


Figure 4.7: Arbitrary spectra illustrating the Shirley background subtraction algorithm.  $E_{1,2}$  are endpoints of the binding energy range considered. The orange and purple cross hatching are the integrals in equation 4.1.

### 4.6.1 Shirley Background Subtraction

For quantitative analysis, peak intensities have to be normalized from scan to scan. The “Shirley” method is one of a variety of ways to subtract the background that was employed for this work [122]. This is an iterative method that accounts for background noise in the detector as well as inelastic scattering processes (see for example Hufner [16] for more information beyond Shirley’s original work).

The intensity  $I$  at each point in the energy range of interest is iteratively rescaled using the following formula:

$$I'(E) = I(E) - [I(E_1) - I(E_2)] \frac{\int_E^{E_2} dE' [I(E') - I(E_2)]}{\int_{E_1}^{E_2} dE' [I(E') - I(E_2)]} \quad (4.1)$$

The numerator and denominator integral expressions are illustrated with orange and purple cross hatching respectively in figure 4.7. Repeating this procedure across different scans makes relative intensity changes in peak height and area quantitatively comparable.

### 4.6.2 Quantifying Peaks

Peak fitting is a field of study in itself. In this instance, we expect that the core-hole lifetime gives a Lorentzian (eqn. 4.3) distribution of energies, while thermal and instrumental signal broadening are described by a Gaussian (eqn. 4.2) distribution. Other inelastic processes may give more complex peak shapes such as the Doniach-Sunjic lineshape [123].

These two peak shape contributions can be convoluted in either product or sum fashion [121]. For this work, the product convolution referred to as a Voigt function was used. The Gaussian line shape is given by:

$$G(x) = G_0 + \frac{A}{\sqrt{\frac{\pi}{2}}W_G} e^{-\frac{2}{W_G^2}(x-x_c)^2} \quad (4.2)$$

The Lorentzian line shape by:

$$L(x) = L_0 + \frac{2A}{\pi} \frac{W_L}{4(x-x_c)^2 + W_L^2} \quad (4.3)$$

and the Voigt function is the convolution of  $G(x)$  and  $L(x)$ :

$$\begin{aligned} V(x; W_G, W_L) &= \int_{-\infty}^{\infty} dt G(t) L(x-t) \\ &= V_0 + A \frac{2 \ln 2}{\pi^{3/2}} \frac{W_L}{W_G^2} \int_{-\infty}^{\infty} dt \frac{e^{-t^2}}{\left(\sqrt{\ln 2} \frac{W_L}{W_G}\right)^2 + \left(\sqrt{4 \ln 2} \frac{x-x_c}{W_G} - t\right)^2} \end{aligned} \quad (4.4)$$

In equations 4.2, 4.3, 4.4  $A$  is the amplitude,  $x_c$  is the peak centre,  $W_G$  and  $W_L$  are the Gaussian and Lorentzian peak widths, respectively, and  $G_0$ ,  $L_0$ ,  $V_0$  are offset terms.

## 4.7 Specific Application of Peak Fitting to Chapter 5 Results

For this experiment, the Gaussian and Lorentzian peak widths ( $W_G$ ,  $W_L$ ) were initially fixed using values from the literature [124], leaving the amplitude and peak

centre as the fit parameters for the fitting of the pristine cleaved  $\text{Bi}_2\text{Se}_3$  samples. For subsequent fits, the Lorentzian contribution remained fixed with the literature value while the Gaussian width was used as an additional fitting parameter. Amplitude change corresponds to the core level signal attenuation as a result of depositing an thicker and thicker overlayer of  $\text{MoO}_3$ , while shifts in energy of the peak centre correspond to changes in the doping. These changes in Fermi level,  $E_F$ , in response to surface transfer doping were monitored using both the Bi 5d and Se 3d core level components of  $\text{Bi}_2\text{Se}_3$  at a photon energy of 100 eV to ensure high surface sensitivity with an overall measurement uncertainty of  $\pm 0.02$  eV, where the error is defined as one standard deviation.

The shift to lower binding energy was the same for both core levels as expected so only the Bi 5d core level results are presented and discussed. In order to determine  $E_F - E_D$  from the Bi 5d<sub>5/2</sub> binding energy we first measured the angle-integrated valence band of a freshly cleaved sample to determine that the Dirac point energy lies  $0.27 \pm 0.05$  eV below the Fermi level (i.e.  $E_F - E_D = 0.27 \pm 0.05$  eV), consistent with ARPES results on lightly doped  $\text{Bi}_2\text{Se}_3$  crystals, where the Fermi level lies just above the bulk conduction band minimum [14].

From this we determine a fixed separation of the Dirac point to the Bi 5d<sub>5/2</sub> orbital of  $24.60 \pm 0.05$  eV. The  $\pm 0.05$  eV error represents a systematic uncertainty in determining  $E_F - E_D$ .

## Chapter 5: Results II: Doping of $\text{Bi}_2\text{Se}_3$ with Molecular Electron Acceptor $\text{MoO}_3$

### 5.1 Overview of Results

Using the XPS techniques outlined in chapter 4, lightly-doped  $\text{Bi}_2\text{Se}_3$  crystals (surface Fermi energy relative to Dirac point  $E_F - E_D \sim 270$  meV) were cleaved in vacuum and exposed to the molecular dopant  $\text{MoO}_3$ . This was done in three aspects: 1) coverage-dependent photoemission spectroscopy of Bi core levels that tracked the Fermi energy shift, 2) Mo core level spectroscopy measured the fraction of charged  $\text{MoO}_3$  molecules, and 3) low-energy cutoff spectroscopy to measure the work function. We find that  $\text{MoO}_3$  is an efficient doping acceptor, saturating at  $E_F - E_D = 100$  meV, well within the topological regime. An induced interface dipole of 0.74 eV was measured, corresponding to  $\sim 10^{13}$  cm $^{-2}$  electrons transferred from the  $\text{Bi}_2\text{Se}_3$ . A doping model based on Fermi-Dirac statistics was implemented to accurately describe the doping behavior including saturation.

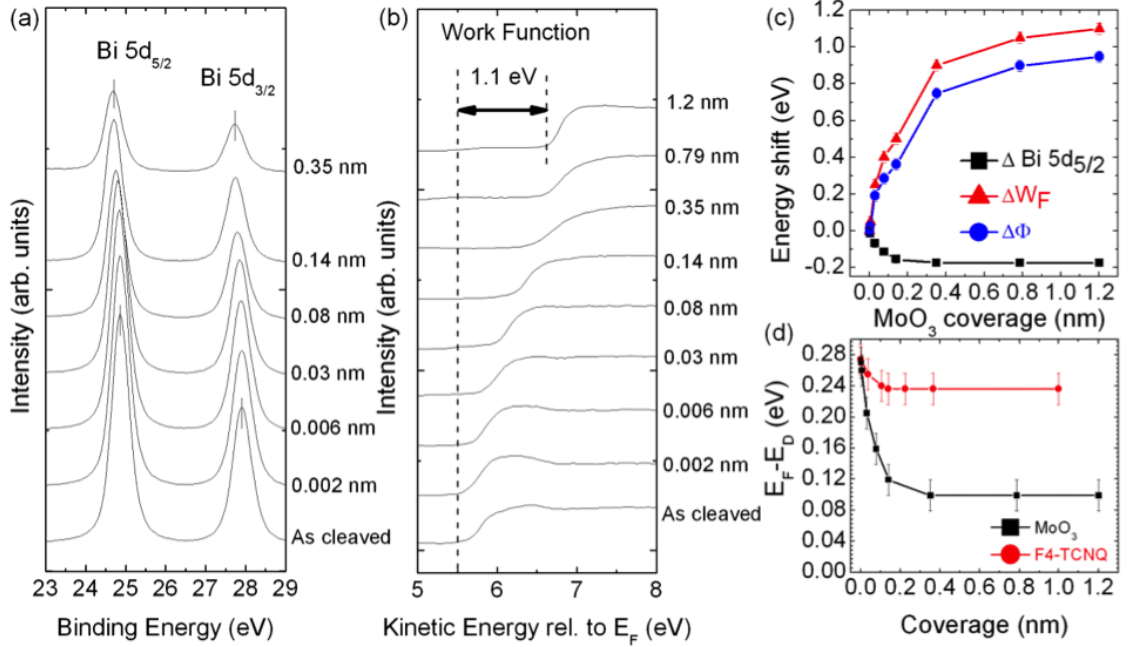


Figure 5.1: Molecular coverage dependent Fermi energy and surface dipole. (a) Bismuth 5d core level spectra taken at  $h\nu = 100$  eV at selected MoO<sub>3</sub> coverages. (b) Secondary electron cut-off as a function MoO<sub>3</sub> coverage. (c) Energy level shifts for the Bi 5d<sub>5/2</sub> core level (black squares), work function (red triangles), and interface dipole (blue circles) as a function of MoO<sub>3</sub> coverage. [Note: Error bars smaller than plot symbol]. (d)  $E_F - E_D$  plotted as a function of MoO<sub>3</sub> coverage (black squares) and F4-TCNQ coverage (red circles).

## 5.2 XPS Results

### 5.2.1 Bi Core Level and Low-energy Cutoff Spectroscopy

In Figure 5.1 the response of the Fermi level shift and work function of *in situ* cleaved Bi<sub>2</sub>Se<sub>3</sub> samples to deposition of MoO<sub>3</sub> are measured using high-resolution surface sensitive photoemission spectroscopy. Figure 5.1(a) plots the evolution of the bismuth 5d core level (at photon energy  $h\nu = 100$  eV) with increasing MoO<sub>3</sub> coverage. Both the Bi 5d<sub>5/2</sub> and Bi 5d<sub>3/2</sub> peak positions shift to lower binding energy (i.e. the Fermi energy moves towards the Dirac point) with increasing coverage, and

reach saturated positions at a coverage of 0.34 nm. In Fig. 5.1(b) the secondary electron cut-off is plotted as a function of MoO<sub>3</sub> coverage. The work function of the freshly cleaved sample is  $5.40 \pm 0.03$  eV, and increases by up to 1.1 eV as the MoO<sub>3</sub> coverage increases. The saturated value of 6.5 eV at a coverage of 0.8 nm is in reasonable agreement with values reported for solid MoO<sub>3</sub> [125], indicating that the low energy cut-off spectra are no longer probing the Bi<sub>2</sub>Se<sub>3</sub>-MoO<sub>3</sub> interface but only the MoO<sub>3</sub> overlayer. In Fig. 5.1(c) the change in Bi 5d<sub>5/2</sub> peak position (black squares), the change in work function,  $\Delta W_F$  (red triangles) and the interface dipole,  $\Delta\phi$  (blue circles), that forms as a result of the surface transfer doping process are plotted as a function of MoO<sub>3</sub> coverage. The binding energy of the Bi core level decreases by  $0.18 \pm 0.02$  eV, consistent with the removal of electrons from the Bi<sub>2</sub>Se<sub>3</sub>. The work function increases by 1.1 eV, indicating the formation of a positive surface dipole that arises due to charge transfer across the Bi<sub>2</sub>Se<sub>3</sub>-MoO<sub>3</sub> interface. The interface dipole is given by  $\Delta\phi = \Delta W_F - \Delta(E_F - E_D)$  where  $E_F - E_D$  is the position of the Fermi level relative to the Dirac point at each coverage plotted in Fig. 5.1(d) (black squares). However, the work function value used to determine the interface dipole is only valid at low coverages (i.e. up until charge transfer saturation), above which it starts to represent the measurement of the molecular overlayer, thus resulting in incorrect determination of the doping induced dipole. At 0.34 nm of MoO<sub>3</sub>, an interface dipole of  $0.73 \pm 0.02$  eV has built up at the Bi<sub>2</sub>Se<sub>3</sub>-MoO<sub>3</sub> interface. In Fig. 5.1(d) we plot  $E_F - E_D$  for both MoO<sub>3</sub> (black squares) and F4-TCNQ (red circles) as a function of respective molecular coverage. In both cases, the as-cleaved samples have  $E_F - E_D = 0.27 \pm 0.05$  eV (i.e. the Fermi level



lies 0.27 eV above the Dirac point energy). However, the response of  $E_F - E_D$  to the two molecules is very different, with only a shift of 40 meV observed for F4-TCNQ, consistent with previous reports of doping to near the conduction band edge [105]. In contrast, for increasing MoO<sub>3</sub> coverage the Fermi level moves towards the Dirac point, reaching a saturated value of  $100 \pm 20$  meV above the Dirac point, placing the Fermi level well within the gap and in the topological regime. This difference could arise due to the significantly larger electron affinity of MoO<sub>3</sub> compared to F4-TCNQ, with values as high as 6.7 eV reported for thick MoO<sub>3</sub> films [125] compared to 5.24 eV for F4-TCNQ [126]. The measured work function of as-cleaved Bi<sub>2</sub>Se<sub>3</sub> is 5.4 eV, placing the lowest unoccupied molecular orbital (LUMO) of F4-TCNQ above the Fermi level of the Bi<sub>2</sub>Se<sub>3</sub> when the vacuum levels are aligned, meaning only a small amount of charge transfer is possible. In contrast, the larger electron affinity of MoO<sub>3</sub> places the LUMO position well below the Fermi level, allowing for charge transfer to take place.

### 5.2.2 Mo Core Level Spectroscopy

Measurement of the Mo core levels revealed three distinct features (figure 5.2(a)): peaks at binding energies of 231.5 eV and 234.7 eV which correspond to the Mo 3d<sub>5/2</sub> and Mo 3d<sub>3/2</sub> states respectively. A third small peak at 228.3 eV is assigned to the 2s core level of sulphur, which we associated with these Bi<sub>2</sub>Se<sub>3</sub> samples. This was confirmed by taking angular-dependent spectra on as-cleaved samples and observing no change in intensity of the S 2s peak; its presence in the

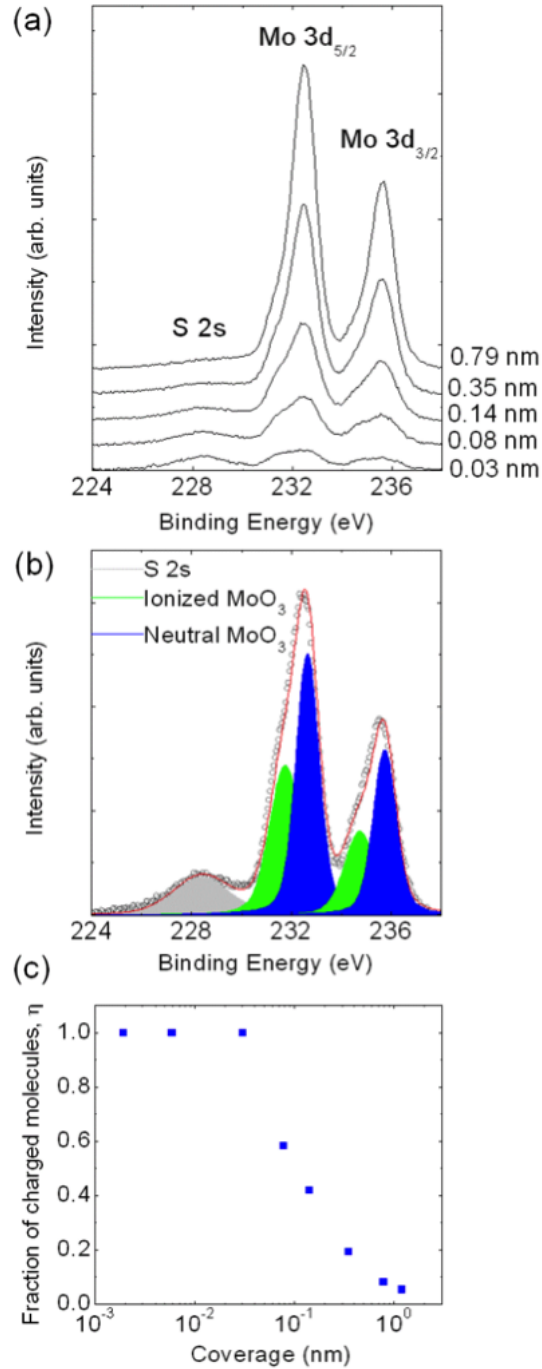


Figure 5.2: Measurement of the fraction of charged molecules using molybdenum core level spectroscopy. (a) Mo 3d core level spectra at selected MoO<sub>3</sub> coverages. (b) MoO<sub>3</sub> core level spectra at a coverage of 0.14 nm with peak fits illustrating the two clear MoO<sub>3</sub> charge states. (c) Relative fraction of ionized MoO<sub>3</sub> as a function of coverage of the two components: doping ( $N_{A-}$ ) and non-doping ( $N_A$ ) that make up the Mo 3d core level of MoO<sub>3</sub> on Bi<sub>2</sub>Se<sub>3</sub>.

data has no bearing on the doping effects being studied. Remarkably, both the Mo  $3d_{5/2}$  and Mo  $3d_{3/2}$  features consist of two distinct peaks clearly separated in binding energy by  $\sim 0.8$  eV and whose relative intensities (branching ratios) are the same for both  $3d_{5/2}$  and  $3d_{3/2}$ . In Fig. 5.2(b) the Mo 3d spectrum taken at 0.14 nm coverage is plotted showing the two distinct components necessary to represent the data. This observation is strong evidence for two chemically different states of the  $\text{MoO}_3$  molecules on the surface of  $\text{Bi}_2\text{Se}_3$  assigned to be neutral ( $N_A$ ) and ionized ( $N_{A-}$ )  $\text{MoO}_3$  acceptors. These two species did not form during evaporation of  $\text{MoO}_3$  (e.g. stoichiometric  $\text{MoO}_3$  and  $\text{MoO}_3$  with oxygen vacancies) because the relative intensities of the two components are strongly dependent on coverage. A similar effect was observed for surface transfer doping hydrogen-terminated diamond with the fluoro-fullerene  $\text{C}_{60}\text{F}_{48}$  where the two species correspond to neutral and singly-ionized  $\text{C}_{60}\text{F}_{48}$  [127]. The Mo 3d lines at lower binding energy are ionized molecules as the extra charge within a  $\text{MoO}_3$  molecule is expected to lower all binding energies due to additional Coulomb repulsion.

The observation of neutral and charged  $\text{MoO}_3$  in the photoemission spectra directly probes the fractional concentration  $\eta = N_{A-} / (N_A + N_{A-})$  of ionized molecules, where  $N_{A-}$  and  $N_A$  are the relative peak areas at each coverage. Figure 5.2(c) shows  $\eta$  plotted as a function of coverage. As expected, at low coverage  $\eta$  is near unity as all the molecules are ionized. At higher coverages  $\eta$  drops as the induced surface dipole retards further charge transfer; there is a transition between mainly ionized  $\text{MoO}_3$  below 0.14 nm to an increasing contribution of non-doping neutral  $\text{MoO}_3$  for higher coverages until the charge transfer process ceases at 0.35

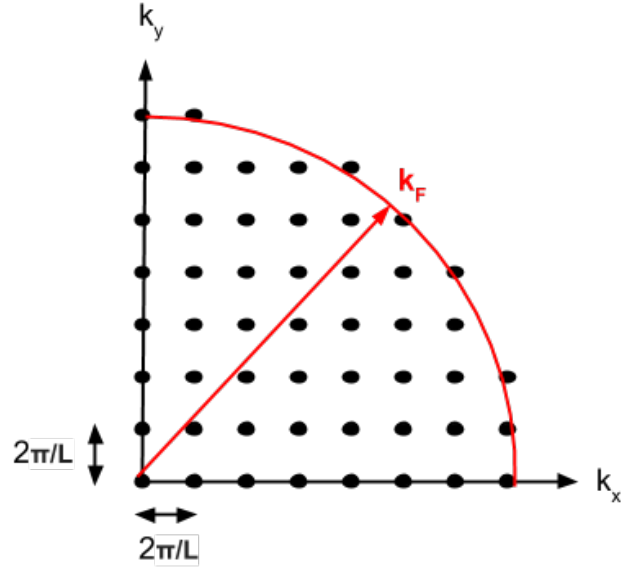


Figure 5.3: How to “count” the two dimensional density of states: 2d momentum space, with black dots representing allowed states, and the Fermi momentum arc indicated in red.

nm. Beyond this coverage additional  $\text{MoO}_3$  remain neutral on the  $\text{Bi}_2\text{Se}_3$  surface.

### 5.2.3 Charge Transfer Model

#### 5.2.3.1 $\text{Bi}_2\text{Se}_3$ Surface States

The simplest way to consider the band structure of  $\text{Bi}_2\text{Se}_3$  is to approximate it as an exactly linear dispersion relation near the Dirac point:

$$E_F - E_D = \hbar\nu_F|\mathbf{k}|$$

In two dimensions, enumerating the density of states is fairly straightforward. Figure 5.3 shows a representation of momentum space, where the allowed states are represented by black dots at  $2\pi/L$  intervals, the condition for the wave function period to match the lattice spacing. The phase space area per state is then  $4\pi^2/L^2$ .

The total phase space area for a given  $k_F$  is  $\pi k_F^2$ . Divide the one by the other to count the number of states:

$$N = \frac{\pi k_F^2}{4\pi^2/L^2} = \frac{k_F^2 L^2}{4\pi} \rightarrow n \equiv \frac{N}{L^2} = \frac{k_F^2}{4\pi}$$

In the case of the topological insulator surface state, the spin-momentum locking means that the momentum states are not doubly spin degenerate, so this expression is not multiplied by two. Substitute this into the original assumption of a linear dispersion to arrive at the desired relationship between carrier density and energy:

$$E_F - E_D = \hbar\nu_F|\mathbf{k}| = \hbar\nu_F\sqrt{4\pi n} \rightarrow n(E_F) = \frac{|E_F - E_D|^2}{\pi(2\hbar\nu_F)^2} \quad (5.1)$$

### 5.2.3.2 MoO<sub>3</sub> Doping

The Fermi energy shift (figure 5.1d) can be understood by interpreting the concentration of ionized MoO<sub>3</sub> (figure 5.2(c)) in terms of the amount of charge removed from the surface. Figure 5.4(a) is a plot of the charged MoO<sub>3</sub> molecules,  $N_{A-}$ , expressed in monolayers (1 ML corresponds to a thickness of 0.65 nm [128]) as a function of  $E_F - E_D$ . If only one electron is transferred per MoO<sub>3</sub> molecule, consistent with the observation of only two types of Mo core-level spectra, and the expectation that strong intermolecular Coulomb interaction should yield positive effective correlation energy,  $N_{A-}$  can be equated with the change in charge density in the Bi<sub>2</sub>Se<sub>3</sub>. In this case, equation 5.1 can be modified to describe the behavior of  $N_{A-}$  with respect to  $E_F - E_D$ :

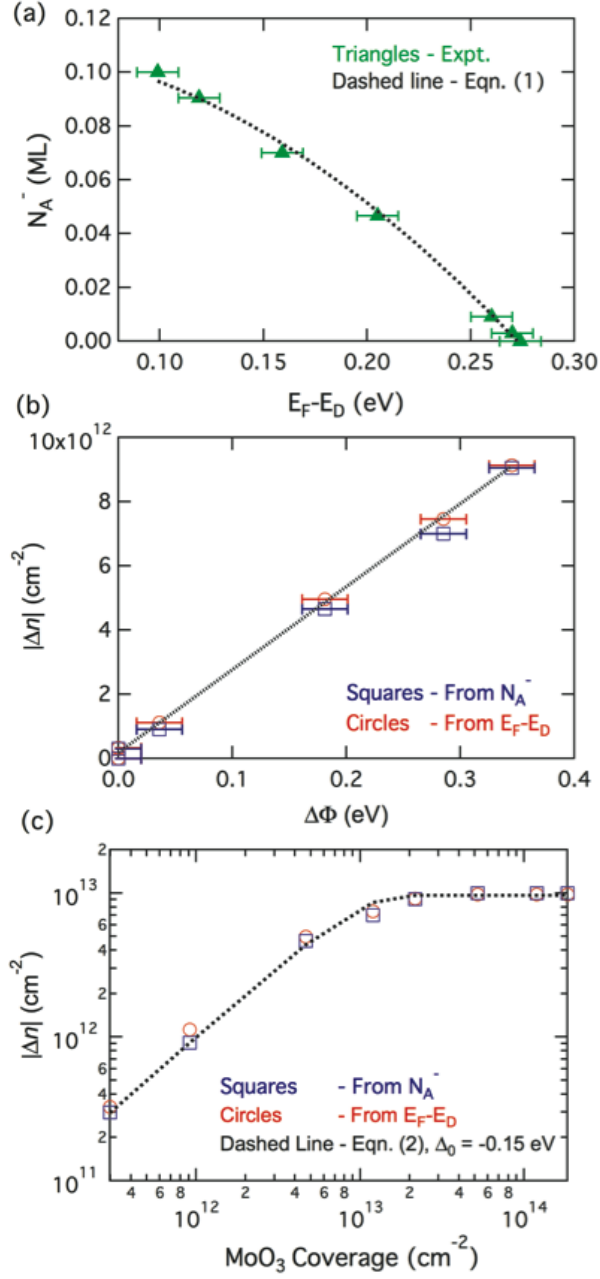


Figure 5.4: Comparison of experimental data and charge transfer model. (a) Number of charged  $\text{MoO}_3$  molecules (monolayer, ML) plotted as a function of  $E_F - E_D$ . The black dashed line is a fit of the data using Eqn. (5.2). (b) Change in carrier density  $|\Delta n|$  as a function of interface dipole  $\Delta\phi$ . (c) Change in carrier density  $|\Delta n|$  as a function of  $\text{MoO}_3$  coverage. In (b-c) blue squares show  $|\Delta n|$  determined from the number of charged  $\text{MoO}_3$  molecules (data in Fig. 5.2c) and red circles show  $|\Delta n|$  calculated from  $E_F - E_D$  assuming a linear surface state dispersion. In (c) the black dashed line corresponds to the change in density calculated using Eqn. (5.3) with initial activation energy of -0.15 eV.

$$N_{A^-} \equiv \frac{n_0}{\rho} - \frac{|E_F - E_D|^2}{\pi\rho(2\hbar\nu_F)^2} \quad (5.2)$$

$\rho$  is the areal density of a monolayer of  $\text{MoO}_3$  molecules per  $\text{cm}^2$ ,  $n_0$  is the initial sheet carrier density of the  $\text{Bi}_2\text{Se}_3$  surface, and  $\nu_F$  is the Fermi velocity of electrons in  $\text{Bi}_2\text{Se}_3$ . [Note that there should be some additional contribution from bulk states in Eqn. (5.2), however the low Fermi energy of the as-cleaved crystals  $E_F - E_D = 270$  meV is near the bulk band edge, so the surface state is expected to dominate the density of states.] The dashed line in Fig. 5.4a is a fit to Eqn. (5.2). The fit does not determine both  $\rho$  and  $\nu_F$ ; however assuming a Fermi velocity of  $\nu_F = 3.5 \times 10^5 \text{ ms}^{-1}$  [14] [129] yields a value for the areal density for 1 ML of,  $\rho = 1.0 \times 10^{14} \text{ MoO}_3 \text{ molecules cm}^{-2}$ , in reasonable agreement with what would be expected from close packing arguments based on the size of the  $\text{MoO}_3$  molecule. Using  $\nu_F = 3.5 \times 10^5 \text{ ms}^{-1}$  and the linear band dispersion, the as-cleaved samples ( $E_F - E_D = 0.270 \text{ eV}$ ), have an initial surface carrier density of  $n_0 = 1.13 \times 10^{13} \text{ cm}^{-2}$ . Upon saturation of the charge transfer process at a coverage of 0.34 nm of  $\text{MoO}_3$  and  $E_F - E_D = 0.10 \text{ eV}$  the density is now  $n = 1.47 \times 10^{12} \text{ cm}^{-2}$ . Thus, almost  $\sim 10^{13} \text{ cm}^{-2}$  of electrons can be removed from the  $\text{Bi}_2\text{Se}_3$  when  $\text{MoO}_3$  is deposited onto the surface.

The change in charge density  $|\Delta n| = |n - n_0|$  can be compared to the interface dipole,  $\Delta\phi$ .  $\Delta n$  is extracted from the experimentally measured  $E_F - E_D$  values (red circles) using  $\nu_F = 3.5 \times 10^5 \text{ ms}^{-1}$ , as well as independently from the number of ionized  $\text{MoO}_3$  molecules (blue squares) using the areal density,  $\rho = 1.0 \times 10^{14}$  obtained from figure 5.4a. The result is shown in figure 5.4b. In both cases,  $\Delta n$

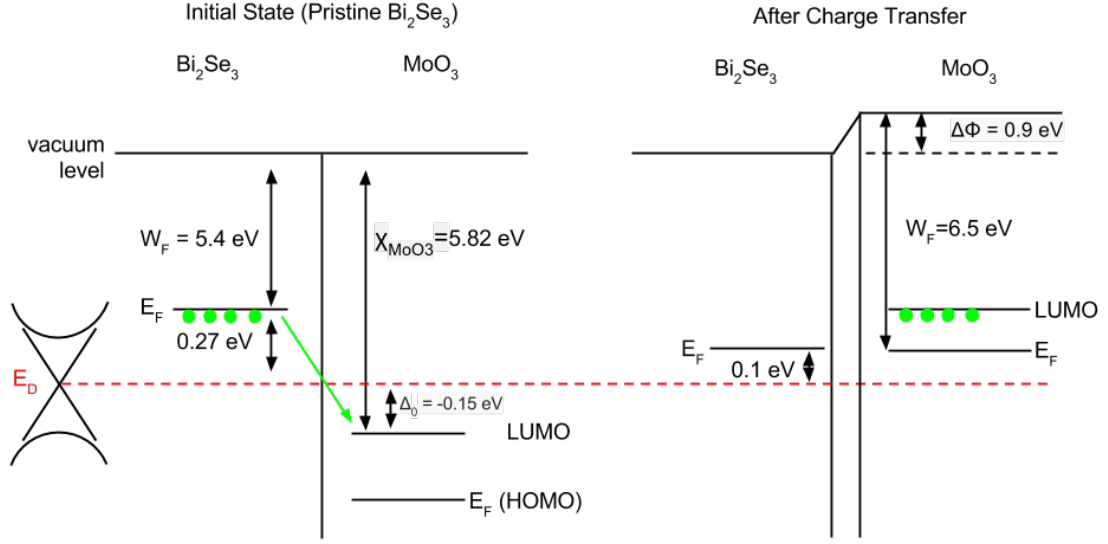


Figure 5.5: Diagram showing the various energies associated with the charge transfer (electrons, in green) from  $\text{Bi}_2\text{Se}_3$  to  $\text{MoO}_3$  molecules. The red dotted line is the Dirac point energy of the  $\text{Bi}_2\text{Se}_3$ . The left diagram shows the Fermi energy ( $E_F$ ) relative to the Dirac point as well as the work function ( $W_F$ ) of pristine  $\text{Bi}_2\text{Se}_3$  prior to doping. The electron affinity of neutral  $\text{MoO}_3$ ,  $\chi_{\text{MoO}_3}$  is also labelled. When charge transfer occurs (right diagram), an interface dipole  $\Delta\phi$  (labelled on right) develops.

is proportional to  $\Delta\phi$ , with the same constant of proportionality, consistent with a simple electrostatic (capacitive) interpretation where  $\Delta n = (\frac{1}{e}C) \Delta\phi$  and  $C$  is the capacitance per area associated with the separation of charge across a charge-free region at the  $\text{Bi}_2\text{Se}_3$ - $\text{MoO}_3$  interface. A linear fit of the data yields  $C = 4.16 \mu\text{F cm}^{-2}$  for the interface capacitance.

The charge transfer mechanism between the  $\text{Bi}_2\text{Se}_3$  and the  $\text{MoO}_3$  overlayer can be thought of as a “band bending” of the  $\text{MoO}_3$  lowest unoccupied molecular orbital (LUMO) and interface dipole formation to match the Fermi levels at the interface (figure 5.5). “Band bending” is strictly speaking a misnomer in this context because the  $\text{MoO}_3$  layer is being considered as discrete molecules as opposed to some kind of crystalline structure. In molecular doping work on diamond, Fermi-Dirac



statistics have been successfully applied to model occupancy of the molecular level into which the electron is transferred [100] [130] [127] [131], in this case the LUMO of MoO<sub>3</sub>.

With all energies referenced to the vacuum level, the LUMO energy corresponds to the electron affinity,  $\chi_{MoO_3}$  of neutral MoO<sub>3</sub> in contact with the Bi<sub>2</sub>Se<sub>3</sub>. Then the doping fraction  $\eta$  can be expressed as:

$$\eta = \frac{N_{A^-}}{N_A + N_{A^-}} = \frac{1}{\frac{1}{g} \exp \left[ \frac{\Delta_0 + \Delta\phi + (E_D - E_F)}{k_B T} \right] + 1} \quad (5.3)$$

Here  $g$  is the degeneracy factor of the MoO<sub>3</sub> LUMO. As above  $N_{A^-}$  is equated with the change in electron density in Bi<sub>2</sub>Se<sub>3</sub>, and  $\Delta_0$  is the initial activation energy before charge transfer has taken place, given by  $\Delta_0 = W_{F,Bi_2Se_3} + (E_F - E_D) - \chi_{MoO_3}$  where  $W_{F,Bi_2Se_3}$  is the work function of Bi<sub>2</sub>Se<sub>3</sub>. Eqn. 5.3 determines the value of  $\Delta_0$  from the coverage dependence of  $\eta$  or  $\Delta n$ .

Fig. 5.4(c) plots the experimentally determined  $|\Delta n|$  as a function of MoO<sub>3</sub> coverage. The red circles represent the change in density determined using the values of  $E_F - E_D$  in Fig. 5.1(d) and Eqn. 5.2. The blue squares are derived making use of the relative intensities of the MoO<sub>3</sub> core level components to determine the number of charged MoO<sub>3</sub> molecules at each coverage (Fig. 5.2c). Overall the agreement between these two independently derived values is very good. The dashed line in Fig. 5.4(c) corresponds to the change in density calculated using Eqn. 5.3 with initial acceptor energy of  $\Delta_0 = -0.15 \pm 0.01$  eV. For the as-cleaved Bi<sub>2</sub>Se<sub>3</sub> the work function was measured to be  $W_F = 5.4$  eV, and the initial position of the Dirac

point relative to the Fermi level,  $E_F - E_D = 0.27$  eV. This yields a value for the electron affinity of neutral  $\text{MoO}_3$ ,  $\chi = 5.82 \pm 0.05$  eV.

### 5.3 Summary of Results

The molecule  $\text{MoO}_3$  is an effective acceptor dopant for  $\text{Bi}_2\text{Se}_3$ . High-resolution synchrotron-based photoelectron spectroscopy measurements indicate that  $\text{MoO}_3$  doping can bring the surface Fermi energy of  $\text{Bi}_2\text{Se}_3$  to  $\sim 100$  meV above the Dirac point, within the so-called ‘topological regime’. There are two distinct molecular charge components in the Mo 3d core level spectra corresponding to charged and uncharged  $\text{MoO}_3$  molecules, allowing a direct measurement of the coverage-dependent charge transfer efficiency. A simple model of charge transfer to a discrete molecular level from the  $\text{Bi}_2\text{Se}_3$  topological surface state accurately describes the coverage dependence of the Fermi energy and interface dipole, and gives an initial activation energy for charge transfer of -0.15 eV.

## Chapter 6: MoO<sub>3</sub> Doping Electrical Transport Results

The XPS study from chapter 5 demonstrated the ability of molecular MoO<sub>3</sub> to put the Fermi level of Bi<sub>2</sub>Se<sub>3</sub> into the topological regime with a low enough initial doping. I extended this work using the film growth and transport techniques from chapters 2 and 3 in pursuit of an air-stable “topological” device.

After a 15 nm Bi<sub>2</sub>Se<sub>3</sub> film had been grown and characterized, the source was changed (while maintaining the sample in vacuum) to deposit an overlayer of MoO<sub>3</sub> while simultaneously measuring the carrier density. After the capping layer was deposited, the same measurement was continued as the sample was vented to ambient.

Fig. 6.1(a) shows the change in carrier density,  $|\Delta n|$  (blue squares) and mobility,  $\mu$  (green circles) of the Bi<sub>2</sub>Se<sub>3</sub> film as a function of MoO<sub>3</sub> coverage. The behavior of the change in carrier density is almost identical to that observed on the as-cleaved Bi<sub>2</sub>Se<sub>3</sub> studied with photoemission: a rapid increase in the change in carrier density at low coverage, followed by a saturation after deposition of  $\sim 0.1$  nm. The overall amount of transferred charge is also very similar at  $\sim 10^{13}$  cm<sup>-2</sup>. The mobility of the film increases to  $\sim 430$  cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup> at low MoO<sub>3</sub> coverages then once the charge transfer process saturates returns to the initial value of  $\sim 390$  cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup>, clear evidence that MoO<sub>3</sub> does not degrade the film quality. A further 100 nm of MoO<sub>3</sub> was

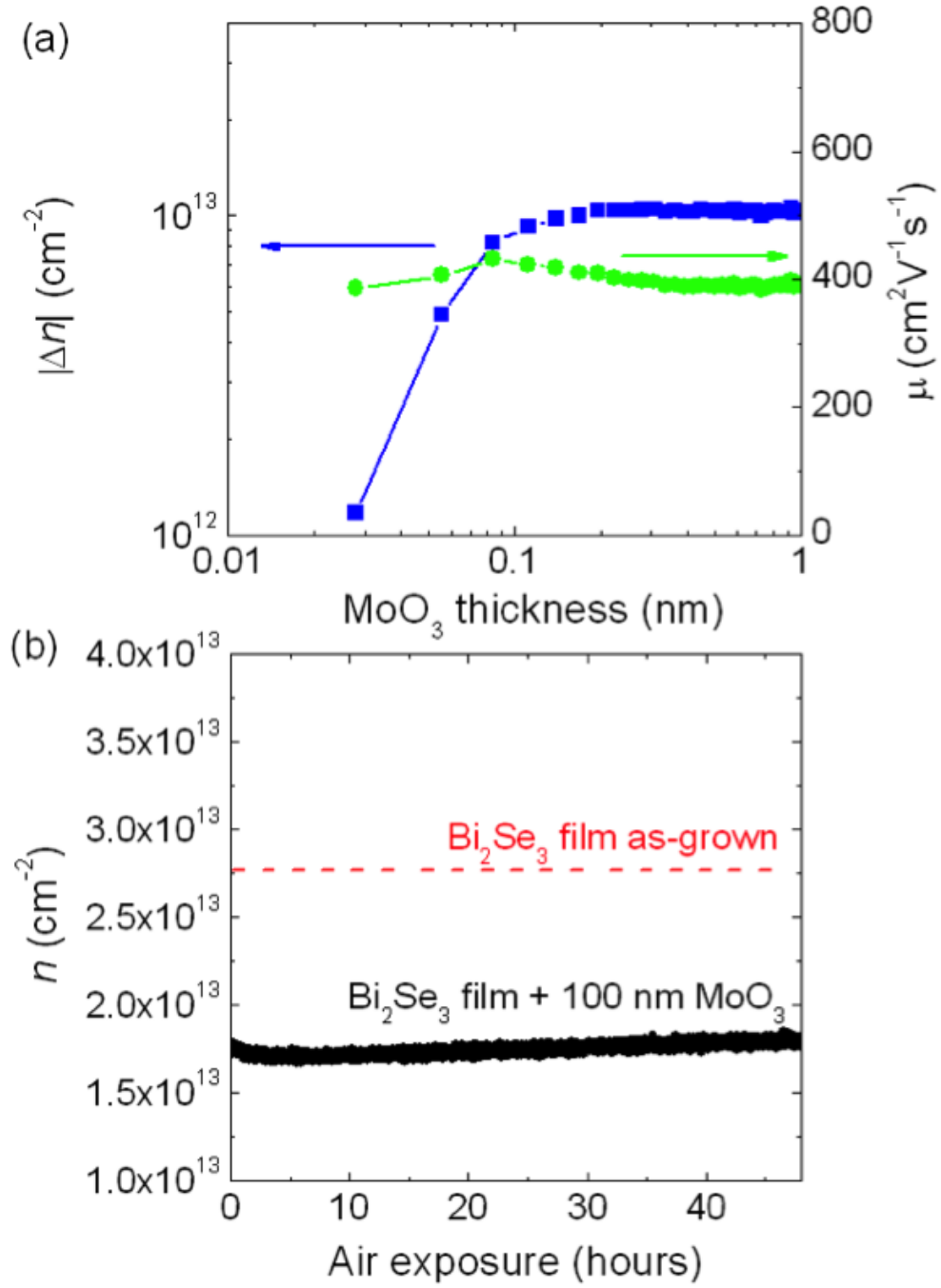


Figure 6.1: Air-stable acceptor doping of a  $\text{Bi}_2\text{Se}_3$  film. (a) Change in carrier density  $|\Delta n|$  determined from Hall effect measurements plotted as a function of  $\text{MoO}_3$  thickness for an MBE-grown  $\text{Bi}_2\text{Se}_3$  thin film. (b) Electron density of the  $\text{Bi}_2\text{Se}_3$  film with 100 nm  $\text{MoO}_3$  passivation layer plotted as a function of time the sample was exposed to ambient conditions (black circles). The red dashed line is a guide to the eye of the density of the as-grown film.

evaporated as a passivation layer, with negligible further change in carrier density. Following the deposition the film was exposed to atmosphere. Fig. 6.1(b) plots the carrier density as a function of exposure time to atmosphere (black circles), with the red dashed line the initial carrier density of the 15 nm film measured after growth but before MoO<sub>3</sub> deposition included as a reference. Over 48 hours the carrier density remains almost unchanged, with only  $\sim 5 \times 10^{11} \text{ cm}^{-2}$  increase in density from the initial value upon atmosphere exposure of  $n = 1.75 \times 10^{13} \text{ cm}^{-2}$ . For  $N \sim 6$  different samples the same slow increase in density on exposure to ambient was observed, despite different initial values of the carrier density. This represents a retention of well over 90% of the initial electron depletion and still  $\sim 10^{13} \text{ cm}^{-2}$  lower than the  $n$ -type doping of the as-grown film. This time period is more than long enough for transfer to another cryostat or vacuum system, or deposition of additional capping layers. The change in carrier density,  $|\Delta n|$ , is quantitatively similar to that observed in photoemission for our Bi<sub>2</sub>Se<sub>3</sub> crystal surfaces.

The thin film Bi<sub>2</sub>Se<sub>3</sub> has higher doping than the single crystal material and hence higher  $E_F - E_D$ . The charge transfer saturation is limited primarily by the doping induced interface dipole, where  $\Delta\phi \sim \Delta n$ , hence we expect similar  $\Delta n$ , in agreement with the observation. This clearly demonstrates the effectiveness of MoO<sub>3</sub> as a passivation layer, protecting the Bi<sub>2</sub>Se<sub>3</sub> from reaction with air, which is well known to increase the  $n$ -type doping [89] [132], and preserving the acceptor doping at the Bi<sub>2</sub>Se<sub>3</sub>-MoO<sub>3</sub> interface.

The exposure of MoO<sub>3</sub> to air results in a large decrease in the work function from 6.8 eV to 5.3 eV [133], which would limit the effectiveness of MoO<sub>3</sub> as an

electron acceptor. However, because the charge transfer process saturates well below 1 nm as seen from Fig. 6.1(a), the additional  $\text{MoO}_3$  layers exposed to air play no role in surface transfer doping and simply act to prevent atmospheric species from reaching and reacting with either the  $\text{MoO}_3$  molecules directly involved in charge transfer or the  $\text{Bi}_2\text{Se}_3$  itself.

A thick (100 nm) passivation layer of  $\text{MoO}_3$  can preserve the acceptor doping from  $\text{MoO}_3$  and prevent degradation of  $\text{Bi}_2\text{Se}_3$  when it is exposed to atmosphere, with the electron density remaining virtually unchanged after two days in ambient. These findings open up new pathways in  $\text{Bi}_2\text{Se}_3$  devices, including electronic devices that not only operate in the topological regime but also are stable in air over prolonged periods of time.

## Chapter 7: Final Summary, Conclusions, and Future Outlook

This project has demonstrated a viable pathway for measuring transport characteristics during epitaxial thin film growth. Contrary to expectation,  $\text{Bi}_2\text{Se}_3$  is  $n$ -doped into the bulk conduction band as it grows in vacuum, before exposure to ambient. Furthermore, the data suggest that the majority of this undesirable doping occurs at the substrate-film and/or film-vacuum interfaces. The molecular doping work was a natural extension of these capabilities, demonstrating the efficacy of molecular  $\text{MoO}_3$  as a  $p$ -dopant and capping layer against ambient exposure.

The achieved resistivity values of  $\sim 1\text{-}3\text{ m}\Omega\cdot\text{cm}$  make these thin film samples very metallic compared to what has been done in creating bulk insulating single crystals ( $\sim 60\text{ m}\Omega\cdot\text{cm}$  at low temperature, [54]), but comparable to other thin film efforts (for example, [115], [72]). The  $\text{MoO}_3$  capping layer experiments showed nearly a factor of two increase in the sample resistivity. It would be interesting to see if this enhancement also happened with bulk crystals of  $\text{Bi}_2\text{Se}_3$  or different topological insulator candidate materials.

The method of the measurement makes it difficult to be more specific about the nature of the doping (if it is the substrate or the vacuum interface, for example). Low temperature, high magnetic field, gated measurements could help to resolve this

ambiguity. The resulting Shubnikov-de Haas oscillations and non-linear magnetoresistance data can be fit to multiple carrier models that have been variously ascribed to bulk, top and bottom surface, and impurity channels (see for example, references in chapter 1.2.1.3, or appendix A), where the regime explored in this thesis has been limited to a single channel analysis. Gating of the top surface independently of the back surface could also providing clarifying data, but that would take substantial changes to the growth technique to avoid additional *ex situ* fabrication steps.

As the field moves forward in measuring spin effects of these topological states, re-engineering the growth stage for devices with contacts on the length scale of the spin coherence length would be an attractive capability. Work is already underway to prepare such a structure with *ex situ* lithography.

This is the first demonstration of growing film by MBE and measuring the *in situ* electrical properties. Beyond bismuth selenide,  $\text{Na}_3\text{Bi}$  has recently been predicted to behave like a Weyl semimetal [134] [135], evincing similar electronic properties to graphene, but in three dimensions instead of two. A glaring experimental complication is that sodium is a highly reactive alkali metal making the standard techniques of film growth and *ex situ* device preparation impossible. This novel, all-vacuum-side technique makes it possible to do electrical characterization of this material without exposure to air. Growth on  $\text{SrTiO}_3$  substrate material would allow for effective back gating of the device at low temperature; using  $\text{Si}:\text{SiO}_2$  as a (back-gateable) substrate is another possibility if growth is possible on an amorphous template.

The experience of this project has already partially informed the design of a



more powerful instrument incorporating aspects of this technique. Adding *in situ* transport to a combined MBE/ scanning tunneling microscopy (STM) system opens up exciting new experimental possibilities. This combination of techniques might be a necessary precursor to making spin-polarized STM measurements on topological insulators, as one example [136].

## Appendix A: Localization and Magnetoresistance

A vast majority of the published low temperature transport data on topological insulators focusses on measurements of resistance versus magnetic field data, specifically weak anti-localization and weak localization. This is one of the best experimental signatures of 2D transport in a topological surface state. These phenomena, in the context of transport studies, were quantitatively explained by Hikami, Larkin and Nagaoka (HLN) [137]. This appendix briefly reviews this effect.

### A.1 Weak Localization

Localization, or Anderson localization, is a quantum mechanical interference phenomenon occurring in low temperature, highly disordered systems. When electrons are in a disordered system, they undergo diffusive transport, scattering off the disorder sites. If the scattering is strong enough, coherent interference induced by the scattering will completely stop the macroscopic diffusion (the electrons are ‘localized’). Figure (A.1) illustrates two such paths, whose constructive interference would lead to the electron getting stuck at A.

Weak localization, intuitively named, is a partial manifestation of this effect: constructive interference of the two paths leads to enhanced backscattering.

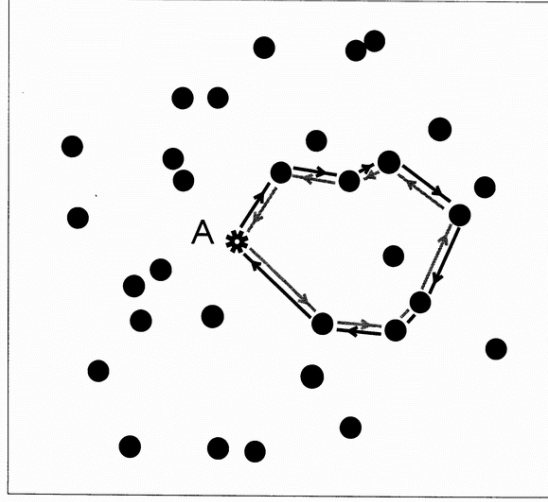


Figure A.1: Illustration of a scattering path of diffusive electron transport in a disordered system (clockwise arrows), with a coherent interfering path (counterclockwise arrows), leading to a tendency for the electron to ‘dwell’ at point A. Figure reproduced from reference [17].

This leads to an increase in resistivity beyond the classical (Drude) expectation. When a magnetic field is applied, the necessary presence of a magnetic potential introduces a phase to the counterpropagating wavefunctions (the Aharonov-Bohm effect), reversing their constructive interference, destroying the localization, lowering the resistivity. Magnetoresistance is therefore a way to experimentally measure weak localization.

## A.2 Weak Anti-Localization

Weak anti-localization is the exact opposite of weak localization. If electron wavefunctions have an additional momentum-dependent phase, the two opposing ‘paths’ will destructively interfere, leading to a suppression of resistivity relative to the classical expectation. The application of a magnetic field will then increase

wavefunction coherence, increasing the resistivity.

The surface states of topological insulators exist as a result of the spin-orbit interaction. This coupling of spin and momentum means that electrons with opposite momentum have opposite spin direction, mathematically manifest as a  $\pi$  Berry phase in their wavefunctions. Weak anti-localization should then be the expected magnetoresistive effect for these materials.

In 1980, Hikami, Larkin and Nagaoka studied the spin-orbit interaction in two dimensions using the renormalization group method, obtaining the following result for the localization effects as a function of the magnetic field in the weak-field limit [137]:

$$\Delta\sigma = -\frac{\alpha e^2}{2\pi^2\hbar} \left[ \ln \frac{1}{\tau_\epsilon a} - \psi\left(\frac{1}{2} + \frac{1}{\tau_\epsilon a}\right) \right] \quad (\text{A.1})$$

where  $\tau_\epsilon$  is the energy relaxation time,  $a = 4DeH/\hbar c$  ( $D$  the diffusion constant,  $H$  the magnetic field).  $\tau_\epsilon a$  is then proportional to the coherence length (owing to its relaxation time and diffusion constant dependence).  $\psi$  is the Digamma function.  $\alpha$  is a parameter whose sign and magnitude determines whether it is weak localization (+1) or anti-localization (-1/2).

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